Glycinonitrile, n-methylene-

Other names:	2-(methyleneamino)acetonitrile
	methyleneaminoacetonitrile
Inchi:	InChI=1S/C3H4N2/c1-5-3-2-4/h1,3H2
InchiKey:	GFZMFCVDDFHSJK-UHFFFAOYSA-N
Formula:	C3H4N2
SMILES:	C=NCC#N
Mol. weight [g/mol]:	68.08
CAS:	109-82-0

Physical Properties

Property code	Value	Unit	Source
hf	150.06	kJ/mol	Joback Method
hvap	35.44	kJ/mol	Joback Method
ie	10.20	eV	NIST Webbook
ie	10.60	eV	NIST Webbook
log10ws	-0.11		Crippen Method
logp	0.211		Crippen Method
mcvol	60.190	ml/mol	McGowan Method
pc	3690.97	kPa	Joback Method
tb	439.32	K	Joback Method
tc	651.30	K	Joback Method
tt	402.15	K	Measurement and correlation of solubility of methyleneaminoacetonitrile in pure and binary solvents and thermodynamic properties of solution
tt	400.19	K	Solubility behaviour and thermodynamic analysis of methyleneaminoacetonitrile in binary (ethanol + water, ethanol + 2-propanol, ethanol + n-butanol) solvents

Sources

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C109820&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Measurement and correlation of solubility of methyleneaminoacetonitrile in pure and binary solvents and thermodynamic properties of solution:	https://www.doi.org/10.1016/j.jct.2019.01.015
Solubility behaviour and thermodynamic analysis of methyleneaminoacetonitrile in binary (ethanol + water, ethanol + 2-propanol, ethanol + n-butanol) solvents:	https://www.doi.org/10.1016/j.jct.2019.06.009
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tt:	Triple Point Temperature

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