

# P-nitro carbanilic acid, n-undecyl ester

<b>Inchi:</b>	InChI=1S/C18H28N2O4/c1-2-3-4-5-6-7-8-9-10-15-24-18(21)19-16-11-13-17(14-12-16)20
<b>InchiKey:</b>	NQKPOXVRXBGLGC-UHFFFAOYSA-N
<b>Formula:</b>	C18H28N2O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)Nc1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	336.43
<b>CAS:</b>	93999-89-4

## Physical Properties

Property code	Value	Unit	Source
gf	94.48	kJ/mol	Joback Method
hf	-391.88	kJ/mol	Joback Method
hfus	55.27	kJ/mol	Joback Method
hvap	90.78	kJ/mol	Joback Method
log10ws	-6.58		Crippen Method
logp	5.674		Crippen Method
mcvol	275.560	ml/mol	McGowan Method
pc	1527.07	kPa	Joback Method
tb	921.20	K	Joback Method
tc	1139.10	K	Joback Method
tf	599.99	K	Joback Method
vc	1.077	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	887.94	J/mol×K	921.20	Joback Method
cpg	902.01	J/mol×K	957.52	Joback Method
cpg	914.95	J/mol×K	993.83	Joback Method
cpg	926.81	J/mol×K	1030.15	Joback Method
cpg	937.64	J/mol×K	1066.47	Joback Method
cpg	947.49	J/mol×K	1102.79	Joback Method
cpg	956.40	J/mol×K	1139.10	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93999894&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93999894&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/16-865-4/P-nitro-carbanilic-acid-n-undecyl-ester.pdf>

Generated by Cheméo on 2024-04-25 19:13:08.804295218 +0000 UTC m=+16361637.724872530.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.