

# Methane, tricyclohexyl-

<b>Other names:</b>	Cyclohexane, 1,1',1''-methylidynetris-Tricyclohexylmethane
<b>Inchi:</b>	InChI=1S/C19H34/c1-4-10-16(11-5-1)19(17-12-6-2-7-13-17)18-14-8-3-9-15-18/h16-19H,
<b>InchiKey:</b>	SQDGPLNZHYWEOB-UHFFFAOYSA-N
<b>Formula:</b>	C19H34
<b>SMILES:</b>	C1CCC(C(C2CCCCC2)C2CCCCC2)CC1
<b>Mol. weight [g/mol]:</b>	262.47
<b>CAS:</b>	1610-24-8

## Physical Properties

Property code	Value	Unit	Source
gf	180.01	kJ/mol	Joback Method
hf	-277.81	kJ/mol	Joback Method
hfus	16.95	kJ/mol	Joback Method
hvap	58.79	kJ/mol	Joback Method
log10ws	-6.49		Crippen Method
logp	6.344		Crippen Method
mvol	245.990	ml/mol	McGowan Method
pc	1708.95	kPa	Joback Method
tb	692.33	K	Joback Method
tc	937.25	K	Joback Method
tf	332.00 ± 2.00	K	NIST Webbook
tf	332.20 ± 2.00	K	NIST Webbook
tf	329.00 ± 3.00	K	NIST Webbook
tf	328.55 ± 0.50	K	NIST Webbook
vc	0.892	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	886.06	J/mol×K	855.61	Joback Method
cpg	928.17	J/mol×K	937.25	Joback Method
cpg	908.12	J/mol×K	896.43	Joback Method
cpg	775.77	J/mol×K	692.33	Joback Method

cpg	806.85	J/mol×K	733.15	Joback Method
cpg	835.52	J/mol×K	773.97	Joback Method
cpg	861.89	J/mol×K	814.79	Joback Method
dvisc	0.0001190	Paxs	692.33	Joback Method
dvisc	0.0002690	Paxs	565.23	Joback Method
dvisc	0.0001717	Paxs	628.78	Joback Method
dvisc	0.0101376	Paxs	311.03	Joback Method
dvisc	0.0025781	Paxs	374.58	Joback Method
dvisc	0.0009754	Paxs	438.13	Joback Method
dvisc	0.0004721	Paxs	501.68	Joback Method
hsubt	117.40	kJ/mol	311.00	NIST Webbook
hvapt	73.30	kJ/mol	516.50	NIST Webbook
hvapt	81.40	kJ/mol	349.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42305e+01
Coeff. B	-5.01608e+03
Coeff. C	-9.20280e+01
Temperature range (K), min.	451.79
Temperature range (K), max.	654.43

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1610248&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1610248&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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