

1-Adamantanecarboxamide, N-(4-methoxyphenyl)-

Inchi:	InChI=1S/C18H23NO2/c1-21-16-4-2-15(3-5-16)19-17(20)18-9-12-6-13(10-18)8-14(7-12)
InchiKey:	DOMQVXNITFPQNL-UHFFFAOYSA-N
Formula:	C18H23NO2
SMILES:	COc1ccc(NC(=O)C23CC4CC(CC(C4)C2)C3)cc1
Mol. weight [g/mol]:	285.38

Physical Properties

Property code	Value	Unit	Source
gf	215.88	kJ/mol	Joback Method
hf	-173.98	kJ/mol	Joback Method
hfus	30.99	kJ/mol	Joback Method
hvap	72.64	kJ/mol	Joback Method
log10ws	-4.28		Crippen Method
logp	3.850		Crippen Method
mcvol	225.560	ml/mol	McGowan Method
pc	2175.46	kPa	Joback Method
tb	789.42	K	Joback Method
tc	1028.47	K	Joback Method
tf	526.34	K	Joback Method
vc	0.855	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	725.64	J/molxK	789.42	Joback Method
cpg	745.46	J/molxK	829.26	Joback Method
cpg	764.58	J/molxK	869.10	Joback Method
cpg	783.28	J/molxK	908.95	Joback Method
cpg	801.81	J/molxK	948.79	Joback Method
cpg	820.42	J/molxK	988.63	Joback Method
cpg	839.39	J/molxK	1028.47	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U307468&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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