

Glutaric acid, hex-4-yn-3-yl hexadecyl ester

Inchi: InChI=1S/C27H48O4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-19-24-30-26(28)22-20-23-25
InchiKey: LCIWRDHFJUWLMB-UHFFFAOYSA-N
Formula: C27H48O4
SMILES: CC#CC(CC)OC(=O)CCCC(=O)OCCCCCCCCCCCCCCCCC
Mol. weight [g/mol]: 436.67

Physical Properties

Property code	Value	Unit	Source
gf	-91.02	kJ/mol	Joback Method
hf	-823.19	kJ/mol	Joback Method
hfus	70.86	kJ/mol	Joback Method
hvap	95.77	kJ/mol	Joback Method
log10ws	-8.76		Crippen Method
logp	7.526		Crippen Method
mcvol	397.570	ml/mol	McGowan Method
pc	793.49	kPa	Joback Method
rinpola	3072.00		NIST Webbook
tb	978.30	K	Joback Method
tc	1201.23	K	Joback Method
tf	629.47	K	Joback Method
vc	1.552	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1354.91	J/molxK	978.30	Joback Method
cpg	1374.81	J/molxK	1015.46	Joback Method
cpg	1393.05	J/molxK	1052.61	Joback Method
cpg	1409.66	J/molxK	1089.77	Joback Method
cpg	1424.69	J/molxK	1126.92	Joback Method
cpg	1438.20	J/molxK	1164.08	Joback Method
cpg	1450.23	J/molxK	1201.23	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U359867&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/16-980-6/Glutaric-acid-hex-4-yn-3-yl-hexadecyl-ester.pdf>

Generated by Cheméo on 2024-04-19 20:19:31.2023793 +0000 UTC m=+15847220.122956615.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.