

# Phthalic acid, 3-methoxybenzyl tridecyl ester

**Inchi:** InChI=1S/C29H40O5/c1-3-4-5-6-7-8-9-10-11-12-15-21-33-28(30)26-19-13-14-20-27(26)2  
**InchiKey:** UPOFRWVKYSFPOF-UHFFFAOYSA-N  
**Formula:** C29H40O5  
**SMILES:** CCCCCCCCCCCCCOC(=O)c1ccccc1C(=O)OCc1ccc(OC)c1  
**Mol. weight [g/mol]:** 468.62

## Physical Properties

Property code	Value	Unit	Source
gf	-173.98	kJ/mol	Joback Method
hf	-813.59	kJ/mol	Joback Method
hfus	64.93	kJ/mol	Joback Method
hvap	106.75	kJ/mol	Joback Method
log10ws	-9.21		Crippen Method
logp	7.520		Crippen Method
mcvol	392.700	ml/mol	McGowan Method
pc	921.06	kPa	Joback Method
rinpol	3506.00		NIST Webbook
tb	1101.24	K	Joback Method
tc	1350.95	K	Joback Method
tf	661.02	K	Joback Method
vc	1.510	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1347.00	J/molxK	1101.24	Joback Method
cpg	1360.24	J/molxK	1142.86	Joback Method
cpg	1371.44	J/molxK	1184.48	Joback Method
cpg	1380.67	J/molxK	1226.10	Joback Method
cpg	1387.99	J/molxK	1267.72	Joback Method
cpg	1393.48	J/molxK	1309.33	Joback Method
cpg	1397.19	J/molxK	1350.95	Joback Method
dvisc	0.0001165	Paxs	661.02	Joback Method
dvisc	0.0000650	Paxs	734.39	Joback Method

dvisc	0.0000403	Paxs	807.76	Joback Method
dvisc	0.0000270	Paxs	881.13	Joback Method
dvisc	0.0000193	Paxs	954.50	Joback Method
dvisc	0.0000145	Paxs	1027.87	Joback Method
dvisc	0.0000113	Paxs	1101.24	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377986&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377986&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/16-996-9/Phthalic-acid-3-methoxybenzyl-tridecyl-ester.pdf>

Generated by Cheméo on 2024-04-27 09:38:03.636519477 +0000 UTC m=+16499932.557096796.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.