# Trichloroethylene

**Other names:**

1,1,2-Trichloroethene
1,1,2-Trichloroethylene
1,1-Dichloro-2-chloroethylene
1,2,2-Trichloroethylene
1-Chloro-2,2-dichloroethylene
Acetylene trichloride
Algylen
Anamenth
Benzinol
Blacosolv
Blancosolv
C2HCl3
CHLORYLEN
Cecolene
Chlorilen
Chlorylea
Chorylen
Circosolv
Crawhaspol
Densinfluat
Distillex DS2
Dow-tri
Dukeron
Ethene, 1,1,2-trichloro-
Ethene, trichloro-
Ethinyl trichloride
Ethylene trichloride
Ethylene, trichloro-
FLUATE
Fleck-flip
Flock FLIP
Gemalgene
Germalgene
Lanadin
Lethurin
NCI-C04546
Narcogen
Narkogen
Narkosoid
Nialk

Perm-A-chlor
Perm-A-clor
Petzinol
Philex
R 1120
RCRA Waste number U228
TCE
TRICHLOROETHENE
Threthylen
Threthylene
Trethylene
Tri
Tri-Clene
Tri-plus
Tri-plus M
Triad
Trial
Triasol
Trichlooretheen
Trichloorethyleen, tri
Trichloraethen
Trichloraethylen, tri
Trichloran
Trichloren
Trichlorethene
Trichlorethylene
Trichlorethylene, tri
Tricloretene
Tricloroetilene
Trielene
Trielin
Trielina
Trieline
Triklone
Triklone N
Trilen
Trilene
Triline
Trimar
Triol
UN 1710
Vestrol
Vitran

| | |
|---|---|
| | Westrosol |
| **Inchi:** | InChI=1S/C2HCl3/c3-1-2(4)5/h1H |
| **InchiKey:** | XSTXAVWGXDQKEL-UHFFFAOYSA-N |
| **Formula:** | C2HCl3 |
| **SMILES:** | ClC=C(Cl)Cl |
| **Mol. weight [g/mol]:** | 131.39 |
| **CAS:** | 79-01-6 |

# Physical Properties

| Property code | Value | Unit | Source |
|---|---|---|---|
| af | 0.2130 | | KDB |
| chl | -956.50 ± 8.40 | kJ/mol | NIST Webbook |
| chl | -947.70 ± 2.90 | kJ/mol | NIST Webbook |
| dm | 0.90 | debye | KDB |
| dvisc | 0.0005510 | Pa×s | Densities and Viscosities of Binary Liquid Mixtures of Trichloroethylene and Tetrachloroethylene with Some Polar and Nonpolar Solvents |
| ea | 0.30 | eV | NIST Webbook |
| ea | 0.40 ± 0.22 | eV | NIST Webbook |
| gf | 19.89 | kJ/mol | KDB |
| gyrad | 3.7590 | | KDB |
| hf | -19.10 ± 3.00 | kJ/mol | NIST Webbook |
| hf | -5.90 | kJ/mol | NIST Webbook |
| hf | -5.86 | kJ/mol | KDB |
| hf | -18.00 ± 2.00 | kJ/mol | NIST Webbook |
| hfl | -41.00 | kJ/mol | NIST Webbook |
| hfl | -53.10 ± 3.00 | kJ/mol | NIST Webbook |
| hfl | -52.00 ± 3.00 | kJ/mol | NIST Webbook |
| hfus | 12.42 | kJ/mol | Joback Method |
| hvap | 34.70 ± 0.42 | kJ/mol | NIST Webbook |
| hvap | 34.50 ± 0.10 | kJ/mol | NIST Webbook |
| hvap | 34.57 ± 0.09 | kJ/mol | NIST Webbook |
| hvap | 34.62 | kJ/mol | NIST Webbook |
| hvap | 33.97 ± 0.13 | kJ/mol | NIST Webbook |
| ie | 9.45 | eV | NIST Webbook |
| ie | 9.46 ± 0.02 | eV | NIST Webbook |
| ie | 9.47 ± 0.01 | eV | NIST Webbook |
| ie | 9.68 | eV | NIST Webbook |
| ie | 9.48 | eV | NIST Webbook |

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|---|---|---|---|
| ie | 9.45 ± 0.01 | eV | NIST Webbook |
| ie | 9.60 | eV | NIST Webbook |
| log10ws | -1.95 | | Aqueous Solubility Prediction Method |
| log10ws | -1.96 | | Estimated Solubility Method |
| logp | 2.502 | | Crippen Method |
| mcvol | 71.460 | ml/mol | McGowan Method |
| nfpaf | %!d(float64=1) | | KDB |
| nfpah | %!d(float64=2) | | KDB |
| nfpas | %!d(float64=1) | | KDB |
| pc | 5020.00 | kPa | KDB |
| rinpol | 686.00 | | NIST Webbook |
| rinpol | 693.00 | | NIST Webbook |
| rinpol | 702.00 | | NIST Webbook |
| rinpol | 690.00 | | NIST Webbook |
| rinpol | 686.00 | | NIST Webbook |
| rinpol | 722.00 | | NIST Webbook |
| rinpol | 672.00 | | NIST Webbook |
| rinpol | 686.00 | | NIST Webbook |
| rinpol | 673.00 | | NIST Webbook |
| rinpol | 680.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 680.00 | | NIST Webbook |
| rinpol | 680.00 | | NIST Webbook |
| rinpol | 701.00 | | NIST Webbook |
| rinpol | 694.00 | | NIST Webbook |
| rinpol | 689.00 | | NIST Webbook |
| rinpol | 681.00 | | NIST Webbook |
| rinpol | 710.00 | | NIST Webbook |
| rinpol | 683.00 | | NIST Webbook |
| rinpol | 690.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 702.00 | | NIST Webbook |
| rinpol | 683.00 | | NIST Webbook |
| rinpol | 702.10 | | NIST Webbook |
| rinpol | 703.00 | | NIST Webbook |
| rinpol | 693.00 | | NIST Webbook |
| rinpol | 690.00 | | NIST Webbook |
| rinpol | 698.00 | | NIST Webbook |
| rinpol | 710.00 | | NIST Webbook |
| rinpol | 686.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |

| | | | |
|---|---|---|---|
| rinpol | 702.00 | | NIST Webbook |
| rinpol | 696.00 | | NIST Webbook |
| rinpol | 675.00 | | NIST Webbook |
| rinpol | 688.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 702.00 | | NIST Webbook |
| rinpol | 681.00 | | NIST Webbook |
| rinpol | 701.00 | | NIST Webbook |
| rinpol | 681.00 | | NIST Webbook |
| rinpol | 689.00 | | NIST Webbook |
| rinpol | 691.00 | | NIST Webbook |
| rinpol | 710.00 | | NIST Webbook |
| rinpol | 686.00 | | NIST Webbook |
| rinpol | 700.00 | | NIST Webbook |
| rinpol | 680.00 | | NIST Webbook |
| rinpol | 694.00 | | NIST Webbook |
| rinpol | 687.00 | | NIST Webbook |
| ripol | 1007.14 | | NIST Webbook |
| ripol | 1010.00 | | NIST Webbook |
| ripol | 1010.00 | | NIST Webbook |
| ripol | 1010.00 | | NIST Webbook |
| ripol | 993.00 | | NIST Webbook |
| ripol | 999.00 | | NIST Webbook |
| ripol | 984.00 | | NIST Webbook |
| ripol | 970.00 | | NIST Webbook |
| ripol | 993.00 | | NIST Webbook |
| ripol | 1028.00 | | NIST Webbook |
| ripol | 1007.14 | | NIST Webbook |
| ripol | 1028.00 | | NIST Webbook |
| ripol | 964.00 | | NIST Webbook |
| ripol | 970.00 | | NIST Webbook |
| ripol | 964.00 | | NIST Webbook |
| ripol | 993.00 | | NIST Webbook |
| ripol | 1010.00 | | NIST Webbook |
| ripol | 1008.00 | | NIST Webbook |
| ripol | 1008.00 | | NIST Webbook |
| ripol | 1019.00 | | NIST Webbook |
| ripol | 1010.00 | | NIST Webbook |
| ripol | 999.00 | | NIST Webbook |
| ripol | 1010.70 | | NIST Webbook |
| ripol | 1012.45 | | NIST Webbook |
| ripol | 993.00 | | NIST Webbook |
| tb | 360.36 | K | KDB |

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|---|---|---|---|
| tb | 360.40 | K | Excess volumes and speeds of sound of mixtures of 1,2-dibromoethane with chlorinated ethanes and ethenes at 303.15 K |
| tb | 360.25 | K | Excess molar enthalpies of dimethylsulfoxide with chloroethanes and chloroethenes at 298.15K |
| tb | 360.25 | K | Vapor-Liquid Equilibria and Excess Molar Enthalpies for N-Methyl-2-pyrrolidone with Chloroethanes and Chloroethenes |
| tb | 360.25 | K | Isobaric Vapor Liquid Equilibrium for Dimethylsulfoxide with Chloroethanes and Chloroethenes |
| tb | 360.40 ± 0.30 | K | NIST Webbook |
| tb | 360.55 ± 0.40 | K | NIST Webbook |
| tb | 360.20 ± 0.40 | K | NIST Webbook |
| tb | 360.20 | K | NIST Webbook |
| tb | 360.00 ± 0.40 | K | NIST Webbook |
| tb | 360.22 ± 0.30 | K | NIST Webbook |
| tb | 360.05 ± 0.20 | K | NIST Webbook |
| tb | 360.05 ± 0.30 | K | NIST Webbook |
| tb | 360.10 ± 0.50 | K | NIST Webbook |
| tb | 360.10 ± 0.50 | K | NIST Webbook |
| tb | 360.25 ± 0.30 | K | NIST Webbook |
| tb | 360.40 ± 0.70 | K | NIST Webbook |
| tb | 360.40 | K | NIST Webbook |
| tc | 571.00 | K | NIST Webbook |
| tc | 544.20 | K | KDB |
| tf | 188.35 ± 0.30 | K | NIST Webbook |
| tf | 188.50 ± 0.20 | K | NIST Webbook |
| tf | 188.40 | K | KDB |
| tf | 188.25 | K | Aqueous Solubility Prediction Method |
| vc | 0.256 | m3/kmol | KDB |
| zc | 0.2840200 | | KDB |
| zra | 0.27 | | KDB |

# Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---|---|---|---|---|

| | | | | |
|---|---|---|---|---|
| cpg | 95.95 | J/mol×K | 500.02 | Joback Method |
| cpg | 100.24 | J/mol×K | 569.29 | Joback Method |
| cpg | 98.20 | J/mol×K | 534.65 | Joback Method |
| cpg | 93.46 | J/mol×K | 465.39 | Joback Method |
| cpg | 90.70 | J/mol×K | 430.76 | Joback Method |
| cpg | 87.67 | J/mol×K | 396.12 | Joback Method |
| cpg | 84.33 | J/mol×K | 361.49 | Joback Method |
| cpl | 128.38 | J/mol×K | 286.17 | Heat capacities of selected chlorohydrocarbons |
| cpl | 130.29 | J/mol×K | 309.14 | Heat capacities of selected chlorohydrocarbons |
| cpl | 129.04 | J/mol×K | 309.14 | Heat capacities of selected chlorohydrocarbons |
| cpl | 129.21 | J/mol×K | 309.14 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.96 | J/mol×K | 301.48 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.54 | J/mol×K | 301.48 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.79 | J/mol×K | 301.48 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.96 | J/mol×K | 296.38 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.46 | J/mol×K | 296.38 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.38 | J/mol×K | 296.38 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.54 | J/mol×K | 296.38 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.46 | J/mol×K | 291.27 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.29 | J/mol×K | 291.27 | Heat capacities of selected chlorohydrocarbons |
| cpl | 130.29 | J/mol×K | 319.35 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.21 | J/mol×K | 291.27 | Heat capacities of selected chlorohydrocarbons |

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|---|---|---|---|---|
| cpl | 130.29 | J/mol×K | 319.35 | Heat capacities of selected chlorohydrocarbons |
| cpl | 131.12 | J/mol×K | 319.35 | Heat capacities of selected chlorohydrocarbons |
| cpl | 130.95 | J/mol×K | 329.56 | Heat capacities of selected chlorohydrocarbons |
| cpl | 131.12 | J/mol×K | 329.56 | Heat capacities of selected chlorohydrocarbons |
| cpl | 128.38 | J/mol×K | 291.27 | Heat capacities of selected chlorohydrocarbons |
| cpl | 120.10 | J/mol×K | 298.00 | NIST Webbook |
| cpl | 124.70 | J/mol×K | 298.00 | NIST Webbook |
| cpl | 126.71 | J/mol×K | 260.64 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.63 | J/mol×K | 260.64 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.63 | J/mol×K | 260.64 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.63 | J/mol×K | 260.64 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.71 | J/mol×K | 265.75 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.80 | J/mol×K | 265.75 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.71 | J/mol×K | 265.75 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.96 | J/mol×K | 265.75 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.71 | J/mol×K | 270.85 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.96 | J/mol×K | 270.85 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.63 | J/mol×K | 270.85 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.04 | J/mol×K | 270.85 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.13 | J/mol×K | 275.96 | Heat capacities of selected chlorohydrocarbons |

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|---|---|---|---|---|
| cpl | 127.04 | J/mol×K | 275.96 | Heat capacities of selected chlorohydrocarbons |
| cpl | 126.96 | J/mol×K | 275.96 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.38 | J/mol×K | 275.96 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.38 | J/mol×K | 281.06 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.38 | J/mol×K | 281.06 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.46 | J/mol×K | 281.06 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.79 | J/mol×K | 281.06 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.96 | J/mol×K | 286.17 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.96 | J/mol×K | 286.17 | Heat capacities of selected chlorohydrocarbons |
| cpl | 127.88 | J/mol×K | 286.17 | Heat capacities of selected chlorohydrocarbons |
| cpl | 129.04 | J/mol×K | 301.48 | Heat capacities of selected chlorohydrocarbons |
| hfust | 8.45 | kJ/mol | 188.50 | NIST Webbook |
| hfust | 8.45 | kJ/mol | 188.50 | NIST Webbook |
| hfust | 8.45 | kJ/mol | 188.50 | NIST Webbook |
| hvapt | 36.20 | kJ/mol | 324.50 | NIST Webbook |
| hvapt | 34.20 | kJ/mol | 328.50 | NIST Webbook |
| hvapt | 31.40 | kJ/mol | 360.40 | NIST Webbook |
| hvapt | 31.38 | kJ/mol | 361.20 | KDB |
| hvapt | 32.13 | kJ/mol | 360.19 | NIST Webbook |
| hvapt | 34.60 | kJ/mol | 354.00 | NIST Webbook |
| hvapt | 35.60 | kJ/mol | 329.00 | NIST Webbook |
| rfi | 1.46980 | | 308.15 | Density, Viscosity, Refractive Index, and Speed of Sound for Binary Mixtures of 1,4-Dioxane with Different Organic Liquids at (298.15, 303.15, and 308.15) K |

| | | | |
|---|---|---|---|
| rfi | 1.47500 | 298.15 | Densities, Excess Molar Volumes, Viscosity, and Refractive Indices of Binary Mixtures of Ethanoic Acid and Trichloroethylene with Dimethylbenzenes at Different Temperatures |
| rfi | 1.47290 | 303.15 | Density, Viscosity, Refractive Index, and Speed of Sound for Binary Mixtures of 1,4-Dioxane with Different Organic Liquids at (298.15, 303.15, and 308.15) K |
| rfi | 1.47500 | 303.15 | Vapor-Liquid Equilibria of Binary Mixtures Formed by Hexan-1-ol with Chloroethanes and Chloroethenes at 95.6 kPa |
| rfi | 1.46900 | 308.15 | Density, Viscosity, Refractive Index, and Speed of Sound in the Binary Mixtures of Tri-n-butylamine + Triethylamine, + Tetrahydrofuran, + Tetradecane, + Tetrachloroethylene, + Pyridine, or + Trichloroethylene at (298.15, 303.15, and 308.15) K |

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|---|---|---|---|
| rfi | 1.47220 | 303.15 | Density, Viscosity, Refractive Index, and Speed of Sound in the Binary Mixtures of Tri-n-butylamine + Triethylamine, + Tetrahydrofuran, + Tetradecane, + Tetrachloroethylene, + Pyridine, or + Trichloroethylene at (298.15, 303.15, and 308.15) K |
| rfi | 1.47520 | 298.15 | Density, Viscosity, Refractive Index, and Speed of Sound in the Binary Mixtures of Tri-n-butylamine + Triethylamine, + Tetrahydrofuran, + Tetradecane, + Tetrachloroethylene, + Pyridine, or + Trichloroethylene at (298.15, 303.15, and 308.15) K |
| rfi | 1.47500 | 298.15 | Bubble-Temperature Measurements on Some Binary Mixtures Formed by Tetrahydrofuran or Amyl Alcohol with Hydrocarbons, Chlorohydrocarbons, or Butanols at (94.6 or 95.8) kPa |
| rfi | 1.47820 | 298.15 | Bubble Temperature Measurements on the Binary Mixtures of n-Heptane or Nitrobenzene or Chlorobenzene with Some Chloroethanes and Chloroethylenes at (94.6 to 95.8) kPa |

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|---|---|---|---|---|
| rfi | 1.47820 | | 298.15 | Activity coefficients and excess Gibbs free energy of some binary mixtures formed by p-cresol at 95.23 kPa |
| rfi | 1.47820 | | 298.15 | Bubble points of the binary mixtures formed by ethylbenzene with some chloroaliphatics and substituted benzenes at p = 94.7 kPa |
| rfi | 1.47820 | | 293.15 | Excess Gibbs energies of selected binary mixtures formed by N,N-dimethyl formamide at 95.5 kPa |
| rfi | 1.47600 | | 298.15 | Density, Viscosity, Refractive Index, and Speed of Sound for Binary Mixtures of 1,4-Dioxane with Different Organic Liquids at (298.15, 303.15, and 308.15) K |
| rfi | 1.47820 | | 293.15 | Bubble points of some binary mixtures formed by o-cresol at 95.75 kPa |
| rhol | 1462.00 | kg/m3 | 293.00 | KDB |
| rhol | 1450.90 | kg/m3 | 303.15 | Excess Volumes and Excess Isentropic Compressibilities of Binary Liquid Mixtures of Trichloroethylene with Esters at 303.15 K |

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|---|---|---|---|---|
| rhol | 1455.90 | kg/m3 | 298.15 | Excess Molar Volumes and Sound Speed in (Phenylacetonitrile + 1,2-Dichloroethane), (Phenylacetonitrile + 1,1,2-Trichloroethane), (Phenylacetonitrile + 1,1,2,2-Tetrachloroethane), (Phenylacetonitrile + Trichloroethene), and (Phenylacetonitrile + Tetrachloroethene) at Temperatures of (303.15, 308.15, and 313.15) K |
| rhol | 1447.50 | kg/m3 | 303.15 | Excess Molar Volumes and Sound Speed in (Phenylacetonitrile + 1,2-Dichloroethane), (Phenylacetonitrile + 1,1,2-Trichloroethane), (Phenylacetonitrile + 1,1,2,2-Tetrachloroethane), (Phenylacetonitrile + Trichloroethene), and (Phenylacetonitrile + Tetrachloroethene) at Temperatures of (303.15, 308.15, and 313.15) K |

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|---|---|---|---|---|
| rhol | 1437.80 | kg/m3 | 308.15 | Excess Molar Volumes and Sound Speed in (Phenylacetonitrile + 1,2-Dichloroethane), (Phenylacetonitrile + 1,1,2-Trichloroethane), (Phenylacetonitrile + 1,1,2,2-Tetrachloroethane), (Phenylacetonitrile + Trichloroethene), and (Phenylacetonitrile + Tetrachloroethene) at Temperatures of (303.15, 308.15, and 313.15) K |
| rhol | 1430.40 | kg/m3 | 313.15 | Excess Molar Volumes and Sound Speed in (Phenylacetonitrile + 1,2-Dichloroethane), (Phenylacetonitrile + 1,1,2-Trichloroethane), (Phenylacetonitrile + 1,1,2,2-Tetrachloroethane), (Phenylacetonitrile + Trichloroethene), and (Phenylacetonitrile + Tetrachloroethene) at Temperatures of (303.15, 308.15, and 313.15) K |
| sfust | 44.80 | J/mol×K | 188.50 | NIST Webbook |
| srf | 0.03 | N/m | 293.20 | KDB |

# Correlations

| Information | Value |
|---|---|
| Property code | pvap |
| Equation | ln(Pvp) = A + B/(T + C) |
| Coeff. A | 1.38164e+01 |
| Coeff. B | -2.66405e+03 |

| | |
|---|---|
| Coeff. C | -7.05700e+01 |
| Temperature range (K), min. | 267.49 |
| Temperature range (K), max. | 383.80 |

| Information | Value |
|---|---|
| Property code | pvap |
| Equation | ln(Pvp) = A + B/T + C*ln(T) + D*T^2 |
| Coeff. A | 3.98315e+01 |
| Coeff. B | -5.14297e+03 |
| Coeff. C | -3.58643e+00 |
| Coeff. D | 1.34232e-06 |
| Temperature range (K), min. | 188.40 |
| Temperature range (K), max. | 571.00 |

# Datasets

## Viscosity, Pa*s

| Temperature, K - Liquid | Pressure, kPa - Liquid | Viscosity, Pa*s - Liquid |
|---|---|---|
| 308.15 | 101.00 | 0.0004600 |

Reference *https://www.doi.org/10.1016/j.jct.2006.06.009*

# Sources

| | |
|---|---|
| **Activity coefficient at infinite dilution measurements for organic solutes (polar and non-polar) in fatty acid methyl esters: Saturated, monounsaturated and polyunsaturated fatty acids: N-Methyl-2-pyrrolidone with Chloroethanes and Chloroethenes.** | https://www.doi.org/10.1016/j.jct.2012.06.009 |
| **Vapor-Liquid Equilibria and Excess Molar Enthalpies for Binary Mixtures of n-Heptane or Nitrobenzene + Chloroethylenes at 298.15 K** | https://www.doi.org/10.1021/je020065c |
| **Density, Viscosity, Refractive Index, and Speed of Sound for Binary Mixtures of Some Chloroethanes and Chloroethenes at (293.15, 298.15, 303.15, and 308.15) K** | https://www.doi.org/10.1021/je0301085 |
| **Excess volumes and speed of sound of mixtures of 1,2-dichloroethane with chloroethenes at 298.15 K** | https://www.doi.org/10.1021/je049609w |
| **Excess molar enthalpies of dimethylsulfoxide with chloroethanes and chloroethenes at 298.15K.** | https://www.doi.org/10.1016/j.jct.2004.11.013 |
| **Excess Molar Volumes and Sound Speed in (Phenylacetonitrile + 1,2-Dichloroethane), (Phenylacetonitrile + Trichloroethylene + 1,1,2-Trichloroethane) + Phenylacetonitrile)** | https://www.doi.org/10.1016/j.tca.2007.08.008 |
| **Densities and Viscosities of Binary Liquid Mixtures of Trichloroethylene + Phenylacetonitrile with Some Polar and Nonpolar Solvents: Estimated Solubility Method** | https://www.doi.org/10.1021/je900570e |
| **Heat Capacities of Selected Chlorohydrocarbons.** | https://www.doi.org/10.1021/je034203p |
| **(Phenylacetonitrile + Trichloroethene), and (Phenylacetonitrile + Tetrachloroethene) at Temperatures of (303.15, 308.15, and 313.15) K:** | http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt |
| | https://www.doi.org/10.1016/j.fluid.2012.09.001 |

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|---|---|
| **KDB:** | https://www.cheric.org/files/research/kdb/mol/mol1719.mol |
| **Aqueous Solubility Prediction Method:** | http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa |
| **Excess Gibbs energies of selected binary mixtures formed by N,N-dimethyl formamide at 95.3 kPa:** | https://www.doi.org/10.1016/j.fluid.2006.08.018 |
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# Legend

| | |
|---|---|
| **af:** | Acentric Factor |
| **chl:** | Standard liquid enthalpy of combustion |
| **cpg:** | Ideal gas heat capacity |
| **cpl:** | Liquid phase heat capacity |
| **dm:** | Dipole Moment |
| **dvisc:** | Dynamic viscosity |
| **ea:** | Electron affinity |
| **gf:** | Standard Gibbs free energy of formation |

| | |
|---|---|
| **gyrad:** | Radius of Gyration |
| **hf:** | Enthalpy of formation at standard conditions |
| **hfl:** | Liquid phase enthalpy of formation at standard conditions |
| **hfus:** | Enthalpy of fusion at standard conditions |
| **hfust:** | Enthalpy of fusion at a given temperature |
| **hvap:** | Enthalpy of vaporization at standard conditions |
| **hvapt:** | Enthalpy of vaporization at a given temperature |
| **ie:** | Ionization energy |
| **log10ws:** | Log10 of Water solubility in mol/l |
| **logp:** | Octanol/Water partition coefficient |
| **mcvol:** | McGowan's characteristic volume |
| **nfpaf:** | NFPA Fire Rating |
| **nfpah:** | NFPA Health Rating |
| **nfpas:** | NFPA Safety Rating |
| **pc:** | Critical Pressure |
| **pvap:** | Vapor pressure |
| **rfi:** | Refractive Index |
| **rhol:** | Liquid Density |
| **rinpol:** | Non-polar retention indices |
| **ripol:** | Polar retention indices |
| **sfust:** | Entropy of fusion at a given temperature |
| **srf:** | Surface Tension |
| **tb:** | Normal Boiling Point Temperature |
| **tc:** | Critical Temperature |
| **tf:** | Normal melting (fusion) point |
| **vc:** | Critical Volume |
| **zc:** | Critical Compressibility |
| **zra:** | Rackett Parameter |