

# Butanedioic acid, ethyl-(1-methylethyl) ester

|                      |                                                                 |
|----------------------|-----------------------------------------------------------------|
| Inchi:               | InChI=1S/C9H16O4/c1-4-12-8(10)5-6-9(11)13-7(2)3/h7H,4-6H2,1-3H3 |
| InchiKey:            | VPEYKSOKDHUUJO-UHFFFAOYSA-N                                     |
| Formula:             | C9H16O4                                                         |
| SMILES:              | CCOC(=O)CCC(=O)OC(C)C                                           |
| Mol. weight [g/mol]: | 188.22                                                          |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -445.38 | kJ/mol  | Joback Method  |
| hf            | -723.97 | kJ/mol  | Joback Method  |
| hfus          | 21.12   | kJ/mol  | Joback Method  |
| hvap          | 53.55   | kJ/mol  | Joback Method  |
| log10ws       | -1.43   |         | Crippen Method |
| logp          | 1.281   |         | Crippen Method |
| mcvol         | 152.550 | ml/mol  | McGowan Method |
| pc            | 2545.61 | kPa     | Joback Method  |
| rinpol        | 1325.00 |         | NIST Webbook   |
| rinpol        | 1324.00 |         | NIST Webbook   |
| rinpol        | 1324.00 |         | NIST Webbook   |
| tb            | 557.46  | K       | Joback Method  |
| tc            | 742.78  | K       | Joback Method  |
| tf            | 320.51  | K       | Joback Method  |
| vc            | 0.582   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 368.40 | J/mol×K | 557.46          | Joback Method |
| cpg           | 381.08 | J/mol×K | 588.35          | Joback Method |
| cpg           | 393.25 | J/mol×K | 619.23          | Joback Method |
| cpg           | 404.90 | J/mol×K | 650.12          | Joback Method |
| cpg           | 416.03 | J/mol×K | 681.01          | Joback Method |
| cpg           | 426.64 | J/mol×K | 711.90          | Joback Method |
| cpg           | 436.72 | J/mol×K | 742.78          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0025423 | Pa×s | 320.51 | Joback Method |
| dvisc | 0.0013123 | Pa×s | 360.00 | Joback Method |
| dvisc | 0.0007719 | Pa×s | 399.49 | Joback Method |
| dvisc | 0.0004996 | Pa×s | 438.99 | Joback Method |
| dvisc | 0.0003474 | Pa×s | 478.48 | Joback Method |
| dvisc | 0.0002553 | Pa×s | 517.97 | Joback Method |
| dvisc | 0.0001960 | Pa×s | 557.46 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R106788&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R106788&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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