

# Acetyl chloride, phenoxy-

<b>Other names:</b>	Phenoxyacetyl chloride Phenyloxyacetyl chloride
<b>Inchi:</b>	InChI=1S/C8H7ClO2/c9-8(10)6-11-7-4-2-1-3-5-7/h1-5H,6H2
<b>InchiKey:</b>	PKUPAJQAJXVUEK-UHFFFAOYSA-N
<b>Formula:</b>	C8H7ClO2
<b>SMILES:</b>	O=C(Cl)COc1ccccc1
<b>Mol. weight [g/mol]:</b>	170.59
<b>CAS:</b>	701-99-5

## Physical Properties

Property code	Value	Unit	Source
gf	-116.96	kJ/mol	Joback Method
hf	-232.46	kJ/mol	Joback Method
hfus	17.50	kJ/mol	Joback Method
hvap	49.22	kJ/mol	Joback Method
log10ws	-1.94		Crippen Method
logp	1.831		Crippen Method
mvol	119.500	ml/mol	McGowan Method
pc	3690.97	kPa	Joback Method
tb	498.70	K	NIST Webbook
tc	747.81	K	Joback Method
tf	308.42	K	Joback Method
vc	0.449	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	238.13	J/molxK	522.84	Joback Method
cpg	285.43	J/molxK	710.32	Joback Method
cpg	277.26	J/molxK	672.82	Joback Method
cpg	268.46	J/molxK	635.33	Joback Method
cpg	259.02	J/molxK	597.83	Joback Method
cpg	248.91	J/molxK	560.34	Joback Method
cpg	292.99	J/molxK	747.81	Joback Method

dvisc	0.0002537	Paxs	522.84	Joback Method
dvisc	0.0003192	Paxs	487.10	Joback Method
dvisc	0.0004165	Paxs	451.37	Joback Method
dvisc	0.0005689	Paxs	415.63	Joback Method
dvisc	0.0008240	Paxs	379.89	Joback Method
dvisc	0.0012890	Paxs	344.16	Joback Method
dvisc	0.0022365	Paxs	308.42	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C701995&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C701995&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/17-074-1/Acetyl-chloride-phenoxy.pdf>

Generated by Cheméo on 2024-04-19 01:34:25.884146258 +0000 UTC m=+15779714.804723573.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.