

Bicyclo[3.3.1]nona-3,7-diene-2,6-dione

Inchi:	InChI=1S/C9H8O2/c10-8-3-1-6-5-7(8)2-4-9(6)11/h1-4,6-7H,5H2
InchiKey:	FEKKORHLKPVNDZ-UHFFFAOYSA-N
Formula:	C9H8O2
SMILES:	O=C1C=CC2CC1C=CC2=O
Mol. weight [g/mol]:	148.16
CAS:	133960-97-1

Physical Properties

Property code	Value	Unit	Source
gf	-75.16	kJ/mol	Joback Method
hf	-261.81	kJ/mol	Joback Method
hfus	10.50	kJ/mol	Joback Method
hvap	45.05	kJ/mol	Joback Method
ie	9.45	eV	NIST Webbook
log10ws	-1.16		Crippen Method
logp	0.887		Crippen Method
mcvol	110.490	ml/mol	McGowan Method
pc	3881.95	kPa	Joback Method
tb	565.57	K	Joback Method
tc	821.71	K	Joback Method
tf	354.47	K	Joback Method
vc	0.415	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	270.26	J/mol×K	565.57	Joback Method
cpg	286.15	J/mol×K	608.26	Joback Method
cpg	301.04	J/mol×K	650.95	Joback Method
cpg	314.91	J/mol×K	693.64	Joback Method
cpg	327.75	J/mol×K	736.33	Joback Method
cpg	339.58	J/mol×K	779.02	Joback Method
cpg	350.37	J/mol×K	821.71	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C133960971&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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