

2,6-Naphthalenedicarboxylic acid, dimethyl ester

Other names:

Dimethyl 2,6-naphthalenedicarboxylate
Dimethyl naphthalene-2,6-dicarboxylate
2,6-Dicarbomethoxynaphthalene
Naphthalene-2,6-dicarboxylic acid, dimethyl ester
Dimethyl ester of 2,6-naphthalenedicarboxylic acid

Inchi: InChI=1S/C14H12O4/c1-17-13(15)11-5-3-10-8-12(14(16)18-2)6-4-9(10)7-11/h3-8H,1-2H**InchiKey:** GYUVMILBYMPKZAZ-UHFFFAOYSA-N**Formula:** C14H12O4**SMILES:** COC(=O)c1ccc2cc(C(=O)OC)ccc2c1**Mol. weight [g/mol]:** 244.24**CAS:** 840-65-3

Physical Properties

Property code	Value	Unit	Source
chs	-6527.60 ± 1.20	kJ/mol	NIST Webbook
gf	-201.04	kJ/mol	Joback Method
hf	-555.10 ± 6.10	kJ/mol	NIST Webbook
hfs	-696.50 ± 1.30	kJ/mol	NIST Webbook
hfus	27.87	kJ/mol	Joback Method
hsub	141.40	kJ/mol	NIST Webbook
hvap	70.31	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	2.413		Crippen Method
mcvol	179.780	ml/mol	McGowan Method
pc	2701.41	kPa	Joback Method
rinsol	2118.00		NIST Webbook
tb	727.92	K	Joback Method
tc	957.77	K	Joback Method
tf	464.50 ± 0.70	K	NIST Webbook
tf	464.40 ± 0.10	K	NIST Webbook
vc	0.681	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	533.23	J/mol×K	957.77	Joback Method
cpg	525.12	J/mol×K	919.46	Joback Method
cpg	516.19	J/mol×K	881.15	Joback Method
cpg	506.41	J/mol×K	842.84	Joback Method
cpg	495.76	J/mol×K	804.54	Joback Method
cpg	484.21	J/mol×K	766.23	Joback Method
cpg	471.72	J/mol×K	727.92	Joback Method
cps	283.80	J/mol×K	298.15	NIST Webbook
dvisc	0.0002177	Paxs	727.92	Joback Method
dvisc	0.0002576	Paxs	685.94	Joback Method
dvisc	0.0003115	Paxs	643.95	Joback Method
dvisc	0.0003867	Paxs	601.97	Joback Method
dvisc	0.0004960	Paxs	559.99	Joback Method
dvisc	0.0006624	Paxs	518.00	Joback Method
dvisc	0.0009310	Paxs	476.02	Joback Method
hfust	53.30	kJ/mol	464.50	NIST Webbook

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C840653&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
cps:	Solid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsub:	Enthalpy of sublimation at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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