

# 3-(«alpha»-Hydroxyethyl)-aniline

<b>Other names:</b>	(3-Aminophenyl)-1-ethanol m-(«alpha»-Hydroxyethyl)aniline 3-Aminophenyl methyl carbinol m-Aminophenylmethylcarbinol m-Amino-«alpha»-methylbenzyl alcohol Benzenemethanol, 3-amino-«alpha»-methyl- m-(1-Hydroxyethyl)aniline Benzyl alcohol, m-amino-«alpha»-methyl- 3-Amino-«alpha»-methylbenzyl alcohol 1-(3-Aminophenyl)ethanol [3-(1-Hydroxyethyl)phenyl]amine NSC 62018 NSC 8392
<b>Inchi:</b>	InChI=1S/C8H11NO/c1-6(10)7-3-2-4-8(9)5-7/h2-6,10H,9H2,1H3
<b>InchiKey:</b>	QPKNDHZQPGLMLCJ-UHFFFAOYSA-N
<b>Formula:</b>	C8H11NO
<b>SMILES:</b>	CC(O)c1cccc(N)c1
<b>Mol. weight [g/mol]:</b>	137.18
<b>CAS:</b>	2454-37-7

## Physical Properties

Property code	Value	Unit	Source
gf	46.45	kJ/mol	Joback Method
hf	-107.11	kJ/mol	Joback Method
hfus	15.89	kJ/mol	Joback Method
hvap	63.27	kJ/mol	Joback Method
log10ws	-1.63		Crippen Method
logp	1.322		Crippen Method
mcvol	115.670	ml/mol	McGowan Method
pc	4368.40	kPa	Joback Method
tb	578.37	K	Joback Method
tc	790.74	K	Joback Method
tf	347.94	K	Joback Method
vc	0.417	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.62	J/mol×K	578.37	Joback Method
cpg	286.22	J/mol×K	613.77	Joback Method
cpg	296.18	J/mol×K	649.16	Joback Method
cpg	305.51	J/mol×K	684.56	Joback Method
cpg	314.24	J/mol×K	719.95	Joback Method
cpg	322.41	J/mol×K	755.35	Joback Method
cpg	330.05	J/mol×K	790.74	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2454377&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2454377&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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