

# Bicyclo[5.1.0]octane

**Inchi:** InChI=1S/C8H14/c1-2-4-7-6-8(7)5-3-1/h7-8H,1-6H2  
**InchiKey:** AKHMHYGTZRJYBS-UHFFFAOYSA-N  
**Formula:** C8H14  
**SMILES:** C1CCC2CC2CC1  
**Mol. weight [g/mol]:** 110.20  
**CAS:** 286-43-1

## Physical Properties

Property code	Value	Unit	Source
chl	-5089.00 ± 1.00	kJ/mol	NIST Webbook
gf	113.78	kJ/mol	Joback Method
hf	-17.00 ± 3.00	kJ/mol	NIST Webbook
hfl	-60.00 ± 2.00	kJ/mol	NIST Webbook
hfus	8.55	kJ/mol	Joback Method
hvap	43.50 ± 0.80	kJ/mol	NIST Webbook
hvap	43.50 ± 0.80	kJ/mol	NIST Webbook
hvap	43.00	kJ/mol	NIST Webbook
log10ws	-2.48		Crippen Method
logp	2.587		Crippen Method
mcvol	101.860	ml/mol	McGowan Method
pc	3560.02	kPa	Joback Method
rinpol	885.00		NIST Webbook
rinpol	885.00		NIST Webbook
rinpol	888.00		NIST Webbook
ripol	1035.00		NIST Webbook
ripol	1045.00		NIST Webbook
tb	414.00 ± 3.00	K	NIST Webbook
tc	614.38	K	Joback Method
tf	208.76	K	Joback Method
vc	0.382	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	198.70	J/molxK	404.46	Joback Method
cpg	217.06	J/molxK	439.45	Joback Method
cpg	234.29	J/molxK	474.43	Joback Method
cpg	250.43	J/molxK	509.42	Joback Method
cpg	265.55	J/molxK	544.41	Joback Method
cpg	279.71	J/molxK	579.39	Joback Method
cpg	292.95	J/molxK	614.38	Joback Method
dvisc	0.0013767	Paxs	208.76	Joback Method
dvisc	0.0010230	Paxs	241.38	Joback Method
dvisc	0.0008158	Paxs	273.99	Joback Method
dvisc	0.0006827	Paxs	306.61	Joback Method
dvisc	0.0005912	Paxs	339.23	Joback Method
dvisc	0.0005251	Paxs	371.84	Joback Method
dvisc	0.0004754	Paxs	404.46	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C286431&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C286431&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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