

# Fumaric acid, hexyl 2,3,5,6-tetrachlorophenyl ester

Inchi:	InChI=1S/C16H16Cl4O4/c1-2-3-4-5-8-23-12(21)6-7-13(22)24-16-14(19)10(17)9-11(18)15
InchiKey:	OFPNLEPZXPVEHX-VOTSOKGWSA-N
Formula:	C16H16Cl4O4
SMILES:	CCCCCOC(=O)C=CC(=O)Oc1c(Cl)c(Cl)cc(Cl)c1Cl
Mol. weight [g/mol]:	414.11

## Physical Properties

Property code	Value	Unit	Source
gf	-277.61	kJ/mol	Joback Method
hf	-618.26	kJ/mol	Joback Method
hfus	52.24	kJ/mol	Joback Method
hvap	91.94	kJ/mol	Joback Method
log10ws	-6.59		Crippen Method
logp	5.885		Crippen Method
mcvol	272.080	ml/mol	McGowan Method
pc	1616.77	kPa	Joback Method
rinsol	2701.00		NIST Webbook
tb	918.54	K	Joback Method
tc	1143.65	K	Joback Method
tf	605.50	K	Joback Method
vc	1.048	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	711.23	J/molxK	918.54	Joback Method
cpg	751.61	J/molxK	1106.13	Joback Method
cpg	745.38	J/molxK	1068.61	Joback Method
cpg	738.25	J/molxK	1031.10	Joback Method
cpg	730.19	J/molxK	993.58	Joback Method
cpg	721.19	J/molxK	956.06	Joback Method
cpg	756.96	J/molxK	1143.65	Joback Method
dvisc	0.0000487	Paxs	918.54	Joback Method
dvisc	0.0000594	Paxs	866.37	Joback Method

dvisc	0.0000744	Paxs	814.19	Joback Method
dvisc	0.0000961	Paxs	762.02	Joback Method
dvisc	0.0001289	Paxs	709.85	Joback Method
dvisc	0.0001811	Paxs	657.67	Joback Method
dvisc	0.0002699	Paxs	605.50	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348252&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348252&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/17-204-6/Fumaric-acid-hexyl-2-3-5-6-tetrachlorophenyl-ester.pdf>

Generated by Cheméo on 2024-04-25 02:05:56.305440736 +0000 UTC m=+16300005.226018049.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.