

# 1,2-Ethanediol dimethanesulphonate

**Other names:** Ethylene glycol, dimethanesulfonate  
Ethylene bis(methanesulfonate)  
Ethylene dimethanesulfonate  
NSC 10716  
1,2-Bis(mesyloxy)ethane  
1,2-Ethanediol dimethanesulfonate  
1,2-Ethanediyl dimethanesulfonate  
Ethane-1,2-diol dimethanesulphonate  
1,2-Ethylene bis(methanesulfonate)  
Dimethylene dimethanesulfonate  
DMSE  
Ethylene methanesulfonate  
EDS  
Methanesulfonic acid, ethylene ester  
NC 429  
Ethane dimethane sulphonate  
1,2-Ethanediol, bismethanesulfonate  
1,2-Ethanediyl methanesulfonate  
Ethylene dimethanesulphonate  
Ethylene-1,2-dimethanesulphonate  
Ethylene glycol dimesylate

**Inchi:** InChI=1S/C4H10O6S2/c1-11(5,6)9-3-4-10-12(2,7)8/h3-4H2,1-2H3

**InchiKey:** QSQFARNGNIZGAW-UHFFFAOYSA-N

**Formula:** C4H10O6S2

**SMILES:** CS(=O)(=O)OCCOS(C)(=O)=O

**Mol. weight [g/mol]:** 218.25

**CAS:** 4672-49-5

## Physical Properties

Property code	Value	Unit	Source
gf	-1164.28	kJ/mol	Joback Method
hf	-1297.03	kJ/mol	Joback Method
hfus	31.25	kJ/mol	Joback Method
hvap	66.59	kJ/mol	Joback Method
log10ws	0.68		Crippen Method
logp	-1.061		Crippen Method

mcvol	135.140	ml/mol	McGowan Method
pc	5687.39	kPa	Joback Method
tb	431.32	K	Joback Method
tc	594.38	K	Joback Method
tf	256.42	K	Joback Method
vc	0.547	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	270.39	J/mol×K	431.32	Joback Method
cpg	279.84	J/mol×K	458.50	Joback Method
cpg	289.12	J/mol×K	485.67	Joback Method
cpg	298.21	J/mol×K	512.85	Joback Method
cpg	307.09	J/mol×K	540.03	Joback Method
cpg	315.72	J/mol×K	567.20	Joback Method
cpg	324.08	J/mol×K	594.38	Joback Method

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C4672495&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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