

# 7-epi-«beta»-Bisabolol

<b>Other names:</b>	(-)-epi-«beta»-bisabolol
<b>Inchi:</b>	InChI=1S/C15H26O/c1-12(2)6-5-11-15(4,16)14-9-7-13(3)8-10-14/h6,14,16H,3,5,7-11H2,
<b>InchiKey:</b>	AOJGKVRVMQAERF-UHFFFAOYSA-N
<b>Formula:</b>	C15H26O
<b>SMILES:</b>	<chem>C=C1CCC(C(C)(O)CCC=C(C)C)CC1</chem>
<b>Mol. weight [g/mol]:</b>	222.37

## Physical Properties

Property code	Value	Unit	Source
gf	90.64	kJ/mol	Joback Method
hf	-267.92	kJ/mol	Joback Method
hfus	20.85	kJ/mol	Joback Method
hvap	64.99	kJ/mol	Joback Method
log10ws	-4.84		Crippen Method
logp	4.230		Crippen Method
mcvol	208.620	ml/mol	McGowan Method
pc	1947.51	kPa	Joback Method
rinpol	1660.00		NIST Webbook
rinpol	1661.00		NIST Webbook
rinpol	1658.00		NIST Webbook
ripol	2163.00		NIST Webbook
tb	654.30	K	Joback Method
tc	851.11	K	Joback Method
tf	324.07	K	Joback Method
vc	0.781	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	588.75	J/molxK	654.30	Joback Method
cpg	606.96	J/molxK	687.10	Joback Method
cpg	624.11	J/molxK	719.90	Joback Method
cpg	640.26	J/molxK	752.71	Joback Method
cpg	655.45	J/molxK	785.51	Joback Method

cpg	669.76	J/mol×K	818.31	Joback Method
cpg	683.24	J/mol×K	851.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R226916&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R226916&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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