

# Dinoseb acetate

**Other names:**

Phenol, 2-(1-methylpropyl)-4,6-dinitro-, acetate  
Acetic acid, 2-(sec-butyl)-4,6-dinitrophenyl ester  
Acetic acid, (2,4-dinitro-6-s-butylphenyl) ester  
Acetic acid, (4,6-dinitro-2-s-butylphenyl) ester  
O-Acetyl-2-sec-butyl-4,6-dinitrophenol  
Aretit  
2-sek.Butyl-4,6-dinitrofenylester kyseliny octove  
2-sec-Butyl-4,6-dinitrophenylacetate  
6-sec-Butyl-2,4-dinitrophenylacetate  
2,4-Dinitro-6-s-butylfenylester kyseliny octove  
2,4-Dinitro-6-sek.butyl-phenylacetat  
4,6-Dinitro-2-s-butylphenyl acetate  
Dinosebe acetate  
HOE 2904  
2-(2-Hydroxy-3,5-dinitrophenyl)butane acetate  
Ivosit  
2-(1-Methylpropyl)-4,6-dinitrophenyl acetate  
Phenol, 2-sec-butyl-4,6-dinitro-, acetate  
Phenol, 2-(1-methylpropyl)-4,6-dinitro-, acetate (ester)  
Phenol, 2-sec-butyl-4,6-dinitro-, acetate (ester)

**Inchi:**

InChI=1S/C12H14N2O6/c1-4-7(2)10-5-9(13(16)17)6-11(14(18)19)12(10)20-8(3)15/h5-7H

**InchiKey:**

RDJTWDKSYLLHRW-UHFFFAOYSA-N

**Formula:**

C12H14N2O6

**SMILES:**

CCC(C)c1cc([N+](=O)[O-])cc([N+](=O)[O-])c1OC(C)=O

**Mol. weight [g/mol]:**

282.25

**CAS:**

2813-95-8

## Physical Properties

Property code	Value	Unit	Source
gf	-31.58	kJ/mol	Joback Method
hf	-360.49	kJ/mol	Joback Method
hfus	41.70	kJ/mol	Joback Method
hvap	88.52	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	2.942		Crippen Method
mcvol	198.460	ml/mol	McGowan Method
pc	2485.07	kPa	Joback Method

rinp	1894.00		NIST Webbook
rinp	1908.00		NIST Webbook
tb	895.11	K	Joback Method
tc	1146.04	K	Joback Method
tf	633.36	K	Joback Method
vc	0.781	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	587.38	J/mol×K	895.11	Joback Method
cpg	597.78	J/mol×K	936.93	Joback Method
cpg	607.02	J/mol×K	978.75	Joback Method
cpg	615.15	J/mol×K	1020.57	Joback Method
cpg	622.18	J/mol×K	1062.40	Joback Method
cpg	628.15	J/mol×K	1104.22	Joback Method
cpg	633.07	J/mol×K	1146.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2813958&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2813958&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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