

# (3-Iodophenyl) methanol, n-pentyl ether

<b>Inchi:</b>	InChI=1S/C12H17IO/c1-2-3-4-8-14-10-11-6-5-7-12(13)9-11/h5-7,9H,2-4,8,10H2,1H3
<b>InchiKey:</b>	WHGFCKHBJHBBKC-UHFFFAOYSA-N
<b>Formula:</b>	C12H17IO
<b>SMILES:</b>	CCCCCOCc1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	304.17

## Physical Properties

Property code	Value	Unit	Source
gf	106.06	kJ/mol	Joback Method
hf	-121.30	kJ/mol	Joback Method
hfus	26.08	kJ/mol	Joback Method
hvap	57.03	kJ/mol	Joback Method
log10ws	-4.68		Crippen Method
logp	3.998		Crippen Method
mcvol	187.870	ml/mol	McGowan Method
pc	2282.77	kPa	Joback Method
rinsol	1760.00		NIST Webbook
tb	621.18	K	Joback Method
tc	847.05	K	Joback Method
tf	344.23	K	Joback Method
vc	0.706	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	429.74	J/molxK	621.18	Joback Method
cpg	496.76	J/molxK	809.41	Joback Method
cpg	485.09	J/molxK	771.76	Joback Method
cpg	472.58	J/molxK	734.12	Joback Method
cpg	459.22	J/molxK	696.47	Joback Method
cpg	444.95	J/molxK	658.83	Joback Method
cpg	507.65	J/molxK	847.05	Joback Method
dvisc	0.0001611	Paxs	621.18	Joback Method
dvisc	0.0002059	Paxs	575.02	Joback Method

dvisc	0.0002747	Paxs	528.86	Joback Method
dvisc	0.0003873	Paxs	482.70	Joback Method
dvisc	0.0005873	Paxs	436.55	Joback Method
dvisc	0.0009825	Paxs	390.39	Joback Method
dvisc	0.0018870	Paxs	344.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374579&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374579&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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