

# Diisopropylethylamine

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>2</sub> CHN(C <sub>2</sub> H <sub>5</sub> )CH(CH <sub>3</sub> ) <sub>2</sub> (i-C <sub>3</sub> H <sub>7</sub> ) <sub>2</sub> (C <sub>2</sub> H <sub>5</sub> )N 1,1'-Dimethyltriethylamine 2-Propanamine, N-ethyl-N-(1-methylethyl)- Bis(1-methylethyl)ethylamine Ethyl-diisopropylamine Hunig's base Hunig's reagent N,N-diisopropylethylamine N-Ethyl-N,N-diisopropylamine N-ethyl-N-(1-methylethyl)-1-methylethanamine N-ethyl-N-isopropylisopropylamine N-ethyl-diisopropylamine NSC 147491 Triethylamine, 1,1'-dimethyl-
<b>Inchi:</b>	InChI=1S/C8H19N/c1-6-9(7(2)3)8(4)5/h7-8H,6H2,1-5H3
<b>InchiKey:</b>	JGFZNNIVVJXRND-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>19</sub> N
<b>SMILES:</b>	CCN(C(C)C)C(C)C
<b>Mol. weight [g/mol]:</b>	129.24
<b>CAS:</b>	7087-68-5

## Physical Properties

Property code	Value	Unit	Source
affp	994.30	kJ/mol	NIST Webbook
basg	963.50	kJ/mol	NIST Webbook
gf	122.38	kJ/mol	Joback Method
hf	-151.48	kJ/mol	Joback Method
hfus	12.45	kJ/mol	Joback Method
hvap	34.67	kJ/mol	Joback Method
log10ws	-1.96		Crippen Method
logp	2.125		Crippen Method
mcvol	133.560	ml/mol	McGowan Method
pc	2571.50	kPa	Joback Method
tb	399.70	K	NIST Webbook
tc	563.76	K	Joback Method
tf	182.39	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	261.03	J/mol×K	394.00	Joback Method
cpg	275.94	J/mol×K	422.29	Joback Method
cpg	290.24	J/mol×K	450.59	Joback Method
cpg	303.95	J/mol×K	478.88	Joback Method
cpg	317.09	J/mol×K	507.17	Joback Method
cpg	329.67	J/mol×K	535.46	Joback Method
cpg	341.72	J/mol×K	563.76	Joback Method
rhoI	742.00	kg/m3	298.15	Physico-Chemical Properties of Non-Newtonian Shear Thickening Diisopropyl-ethylammonium-Based Protic Ionic Liquids and Their Mixtures with Water and Acetonitrile

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7087685&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Physico-Chemical Properties of Non-Newtonian Shear Thickening Diisopropyl-ethylammonium-Based Protic Ionic Liquids and Their Mixtures with Water and Acetonitrile:**

<https://www.doi.org/10.1021/je101191e>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rho:</b>	Liquid Density
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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