

# 1-Butanol, 3-methyl-, carbonate (2:1)

|                      |                                                                        |
|----------------------|------------------------------------------------------------------------|
| Other names:         | Di-iso-amyl carbonate<br>diisopentyl carbonate                         |
| Inchi:               | InChI=1S/C11H22O3/c1-9(2)5-7-13-11(12)14-8-6-10(3)4/h9-10H,5-8H2,1-4H3 |
| InchiKey:            | DWWPOAYNAWHBKI-UHFFFAOYSA-N                                            |
| Formula:             | C11H22O3                                                               |
| SMILES:              | CC(C)CCOC(=O)OCCC(C)C                                                  |
| Mol. weight [g/mol]: | 202.29                                                                 |
| CAS:                 | 2050-95-5                                                              |

## Physical Properties

| Property code | Value         | Unit    | Source         |
|---------------|---------------|---------|----------------|
| gf            | -302.06       | kJ/mol  | Joback Method  |
| hf            | -657.95       | kJ/mol  | Joback Method  |
| hfus          | 21.18         | kJ/mol  | Joback Method  |
| hvap          | 50.87         | kJ/mol  | Joback Method  |
| log10ws       | -2.87         |         | Crippen Method |
| logp          | 3.232         |         | Crippen Method |
| mcvol         | 179.160       | ml/mol  | McGowan Method |
| pc            | 2012.70       | kPa     | Joback Method  |
| tb            | 505.35 ± 0.40 | K       | NIST Webbook   |
| tc            | 725.84        | K       | Joback Method  |
| tf            | 278.12        | K       | Joback Method  |
| vc            | 0.681         | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 442.83 | J/mol×K | 548.91          | Joback Method |
| cpg           | 458.24 | J/mol×K | 578.40          | Joback Method |
| cpg           | 473.06 | J/mol×K | 607.89          | Joback Method |
| cpg           | 487.29 | J/mol×K | 637.38          | Joback Method |
| cpg           | 500.93 | J/mol×K | 666.86          | Joback Method |
| cpg           | 513.98 | J/mol×K | 696.35          | Joback Method |
| cpg           | 526.44 | J/mol×K | 725.84          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0043534 | Pa×s | 278.12 | Joback Method |
| dvisc | 0.0016670 | Pa×s | 323.25 | Joback Method |
| dvisc | 0.0008076 | Pa×s | 368.38 | Joback Method |
| dvisc | 0.0004583 | Pa×s | 413.52 | Joback Method |
| dvisc | 0.0002908 | Pa×s | 458.65 | Joback Method |
| dvisc | 0.0002001 | Pa×s | 503.78 | Joback Method |
| dvisc | 0.0001465 | Pa×s | 548.91 | Joback Method |

## Sources

|                        |                                                                                                                                             |
|------------------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2050955&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2050955&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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