

# Benzoyl chloride, 2-fluoro-

<b>Other names:</b>	o-Fluorobenzoyl chloride 2-Fluorobenzoyl chloride ortho-Fluorobenzoyl chloride Benzoyl chloride, o-fluoro-
<b>Inchi:</b>	InChI=1S/C7H4ClFO/c8-7(10)5-3-1-2-4-6(5)9/h1-4H
<b>InchiKey:</b>	RAAGZOYMEQDCTD-UHFFFAOYSA-N
<b>Formula:</b>	C7H4ClFO
<b>SMILES:</b>	O=C(Cl)c1cccc1F
<b>Mol. weight [g/mol]:</b>	158.56
<b>CAS:</b>	393-52-2

## Physical Properties

Property code	Value	Unit	Source
gf	-224.82	kJ/mol	Joback Method
hf	-287.18	kJ/mol	Joback Method
hfus	16.41	kJ/mol	Joback Method
hvap	44.43	kJ/mol	Joback Method
log10ws	-2.69		Crippen Method
logp	2.205		Crippen Method
mcvol	101.310	ml/mol	McGowan Method
pc	3940.66	kPa	Joback Method
tb	481.79	K	Joback Method
tc	702.19	K	Joback Method
tf	288.03	K	Joback Method
vc	0.393	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.28	J/molxK	481.79	Joback Method
cpg	193.11	J/molxK	518.52	Joback Method
cpg	201.35	J/molxK	555.26	Joback Method
cpg	209.03	J/molxK	591.99	Joback Method
cpg	216.17	J/molxK	628.72	Joback Method

cpg	222.79	J/mol×K	665.46	Joback Method
cpg	228.92	J/mol×K	702.19	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	364.00 ± 1.00	K	2.00	NIST Webbook
tbrp	364.00 ± 1.00	K	2.00	NIST Webbook
tbrp	364.20	K	2.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C393522&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C393522&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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