

# Methanesulfenyl chloride, trichloro-

<b>Other names:</b>	(Trichloromethane)sulfenyl chloride (Trichloromethyl)sulfenyl chloride Perchloromethyl mercaptan Clairsit Mercaptan methylique perchlore Perchlormethylmerkaptan PCM Rcra waste number P118 Trichloromethanesulphenyl chloride Trichloromethylsulphenyl chloride UN 1670 NSC 66404 Perchloromethanethiol
<b>Inchi:</b>	InChI=1S/CCI4S/c2-1(3,4)6-5
<b>InchiKey:</b>	RYFZYYUIAZYQLC-UHFFFAOYSA-N
<b>Formula:</b>	CCI4S
<b>SMILES:</b>	CISC(CI)(CI)CI
<b>Mol. weight [g/mol]:</b>	185.89
<b>CAS:</b>	594-42-3

## Physical Properties

Property code	Value	Unit	Source
gf	-54.22	kJ/mol	Joback Method
hf	-93.81	kJ/mol	Joback Method
hfl	-75.70	kJ/mol	NIST Webbook
hfus	11.85	kJ/mol	Joback Method
hvap	40.88	kJ/mol	Joback Method
log10ws	-3.32		Crippen Method
logp	3.201		Crippen Method
mcvol	90.260	ml/mol	McGowan Method
pc	4917.69	kPa	Joback Method
tb	437.55	K	Joback Method
tc	684.95	K	Joback Method
tf	257.53	K	Joback Method
vc	0.331	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	116.50	J/mol×K	437.55	Joback Method
cpg	119.54	J/mol×K	478.78	Joback Method
cpg	122.16	J/mol×K	520.02	Joback Method
cpg	124.37	J/mol×K	561.25	Joback Method
cpg	126.21	J/mol×K	602.48	Joback Method
cpg	127.73	J/mol×K	643.72	Joback Method
cpg	128.95	J/mol×K	684.95	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C594423&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C594423&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/17-481-9/Methanesulfenyl-chloride-trichloro.pdf>

Generated by Cheméo on 2024-04-20 07:08:30.428263664 +0000 UTC m=+15886159.348840979.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.