

3-Nitro-4-methylbenzaldehyde

Other names:	Benzaldehyde, 4-methyl-3-nitro- 4-Methyl-3-nitrobenzaldehyde 3-nitro-p-tolualdehyde
Inchi:	InChI=1S/C8H7NO3/c1-6-2-3-7(5-10)4-8(6)9(11)12/h2-5H,1H3
InchiKey:	KHWGAWBXQOKXIJ-UHFFFAOYSA-N
Formula:	C8H7NO3
SMILES:	<chem>Cc1ccc(C=O)cc1[N+](=O)[O-]</chem>
Mol. weight [g/mol]:	165.15
CAS:	31680-07-6

Physical Properties

Property code	Value	Unit	Source
gf	45.66	kJ/mol	Joback Method
hf	-91.20	kJ/mol	Joback Method
hfus	23.39	kJ/mol	Joback Method
hvap	60.31	kJ/mol	Joback Method
log10ws	-2.84		Crippen Method
logp	1.716		Crippen Method
mcvol	118.810	ml/mol	McGowan Method
pc	3911.14	kPa	Joback Method
tb	619.58	K	Joback Method
tc	866.72	K	Joback Method
tf	416.99	K	Joback Method
vc	0.474	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.16	J/molxK	619.58	Joback Method
cpg	286.40	J/molxK	660.77	Joback Method
cpg	295.87	J/molxK	701.96	Joback Method
cpg	304.60	J/molxK	743.15	Joback Method
cpg	312.63	J/molxK	784.34	Joback Method
cpg	319.99	J/molxK	825.53	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C31680076&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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