

# Propanoic acid, 2,2-dimethyl-, 4-nitrophenyl ester

<b>Other names:</b>	2,2-Dimethylpropanoic acid, 4-nitrophenyl ester p-nitrophenyl pivalate
<b>Inchi:</b>	InChI=1S/C11H13NO4/c1-11(2,3)10(13)16-9-6-4-8(5-7-9)12(14)15/h4-7H,1-3H3
<b>InchiKey:</b>	QADVJDGFQGNSIF-UHFFFAOYSA-N
<b>Formula:</b>	C11H13NO4
<b>SMILES:</b>	CC(C)(C)C(=O)Oc1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	223.23
<b>CAS:</b>	4195-17-9

## Physical Properties

Property code	Value	Unit	Source
gf	-51.01	kJ/mol	Joback Method
hf	-309.62	kJ/mol	Joback Method
hfus	24.63	kJ/mol	Joback Method
hvap	67.47	kJ/mol	Joback Method
log10ws	-3.45		Crippen Method
logp	2.546		Crippen Method
mcvol	166.950	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpol	1602.00		NIST Webbook
tb	707.64	K	Joback Method
tc	953.76	K	Joback Method
tf	470.86	K	Joback Method
vc	0.638	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	448.61	J/molxK	707.64	Joback Method
cpg	461.74	J/molxK	748.66	Joback Method
cpg	473.77	J/molxK	789.68	Joback Method
cpg	484.76	J/molxK	830.70	Joback Method
cpg	494.78	J/molxK	871.72	Joback Method
cpg	503.88	J/molxK	912.74	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4195179&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4195179&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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