

Decanedioic acid, dibutyl ester

Other names:

Bis(n-butyl) sebacate
Butyl sebacate
Decanedioic acid, 1,10-dibutyl ester
Di(n-butyl) sebacate
Dibutyl decanedioate
Dibutyl sebacate
Dibutyl sebacinate
Dibutylester kyseliny sebakove
Kodaflex DBS
Monoplex dbs
NSC 3893
PX 404
Plasthall DBS
Polycizer DBS
Reomol DBS
Sebacic acid, dibutyl ester
Staflex DBS
Uniflex DBS
decandioic acid, dibutyl ester
n-Butyl sebacate

Inchi:

InChI=1S/C18H34O4/c1-3-5-15-21-17(19)13-11-9-7-8-10-12-14-18(20)22-16-6-4-2/h3-16

InchiKey:

PYGXAGIECVVIOZ-UHFFFAOYSA-N

Formula:

C18H34O4

SMILES:

CCCCOC(=O)CCCCCCCCC(=O)OCCCC

Mol. weight [g/mol]:

314.46

CAS:

109-43-3

Physical Properties

| Property code | Value | Unit | Source |
|---------------|--------------|--------|--------------------------------------|
| gf | -367.16 | kJ/mol | Joback Method |
| hf | -904.45 | kJ/mol | Joback Method |
| hfus | 47.95 | kJ/mol | Joback Method |
| hvap | 91.80 ± 3.20 | kJ/mol | NIST Webbook |
| log10ws | -3.72 | | Aqueous Solubility Prediction Method |
| log10ws | -3.90 | | Estimated Solubility Method |

| | | | |
|--------|---------|----------------------|--------------------------------------|
| logp | 4.794 | | Crippen Method |
| mvol | 279.360 | ml/mol | McGowan Method |
| pc | 1227.70 | kPa | Joback Method |
| rinpol | 2180.10 | | NIST Webbook |
| rinpol | 2137.00 | | NIST Webbook |
| rinpol | 2137.00 | | NIST Webbook |
| rinpol | 2169.00 | | NIST Webbook |
| rinpol | 2140.00 | | NIST Webbook |
| rinpol | 2137.00 | | NIST Webbook |
| tb | 763.82 | K | Joback Method |
| tc | 942.53 | K | Joback Method |
| tf | 262.65 | K | Aqueous Solubility Prediction Method |
| vc | 1.091 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 849.49 | J/molxK | 763.82 | Joback Method |
| cpg | 866.90 | J/molxK | 793.61 | Joback Method |
| cpg | 883.40 | J/molxK | 823.39 | Joback Method |
| cpg | 898.98 | J/molxK | 853.18 | Joback Method |
| cpg | 913.68 | J/molxK | 882.96 | Joback Method |
| cpg | 927.49 | J/molxK | 912.75 | Joback Method |
| cpg | 940.44 | J/molxK | 942.53 | Joback Method |
| cpl | 619.00 | J/molxK | 312.00 | NIST Webbook |
| dvisc | 0.0000707 | Paxs | 763.82 | Joback Method |
| dvisc | 0.0005169 | Paxs | 491.42 | Joback Method |
| dvisc | 0.0010365 | Paxs | 436.94 | Joback Method |
| dvisc | 0.0001878 | Paxs | 600.38 | Joback Method |
| dvisc | 0.0001284 | Paxs | 654.86 | Joback Method |
| dvisc | 0.0000931 | Paxs | 709.34 | Joback Method |
| dvisc | 0.0002962 | Paxs | 545.90 | Joback Method |
| hvapt | 106.40 | kJ/mol | 500.00 | NIST Webbook |
| hvapt | 87.60 | kJ/mol | 466.00 | NIST Webbook |
| hvapt | 88.10 | kJ/mol | 327.00 | NIST Webbook |
| hvapt | 94.30 | kJ/mol | 460.50 | NIST Webbook |

rfi

1.44200

298.10

Isobaric vapor
liquid equilibria of
the binary
system: Maleic
anhydride +
di-n-butylsebacate
at 2.67, 5.33 and
8.00 kPa

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.47404e+01 |
| Coeff. B | -5.63216e+03 |
| Coeff. C | -9.08550e+01 |
| Temperature range (K), min. | 480.55 |
| Temperature range (K), max. | 688.18 |

Sources

| | |
|---|---|
| Isobaric vapor liquid equilibria of the binary system: Maleic anhydride + di-n-butylsebacate at 2.67, 5.33 and 8.00 kPa: NIST Webbook: | https://www.doi.org/10.1016/j.fluid.2008.08.001 |
| Estimated Solubility Method: | http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx |
| Crippen Method: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C109433&Units=SI |
| Liquid-Liquid Equilibria of Formic Acid and Furfural in a Biphasic Aqueous-Organic System: Optimization of Solvent and Amine Extractant: Joback Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| McGowan Method: | https://www.doi.org/10.1021/acs.jced.8b00335 |
| Estimated Solubility Method: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| | https://en.wikipedia.org/wiki/Joback_method |
| | http://link.springer.com/article/10.1007/BF02311772 |
| | http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt |

Legend

| | |
|---------------------|---|
| cp _g : | Ideal gas heat capacity |
| cp _l : | Liquid phase heat capacity |
| dv _{isc} : | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |

| | |
|-----------------|---|
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rfi: | Refractive Index |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.cheméo.com/cid/17-712-2/Decanedioic-acid-dibutyl-ester.pdf>

Generated by Cheméo on 2024-04-25 15:17:02.531636091 +0000 UTC m=+16347471.452213409.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.