

# Docosanoic acid, 1,2,3-propanetriyl ester

<b>Other names:</b>	Docosanoïn, tri- Glyceryl behenate Tribehenin Glyceryl tribehenate Syncrowax HR-C propane-1,2,3-triyl tridocosanoate
<b>Inchi:</b>	InChI=1S/C69H134O6/c1-4-7-10-13-16-19-22-25-28-31-34-37-40-43-46-49-52-55-58-61
<b>InchiKey:</b>	DMBUODUULYCPAK-UHFFFAOYSA-N
<b>Formula:</b>	C69H134O6
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCCCC(=O)OCC(COC(=O)CCCCCCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	1059.80
<b>CAS:</b>	18641-57-1

## Physical Properties

Property code	Value	Unit	Source
chs	-43208.20 ± 1.80	kJ/mol	NIST Webbook
gf	-174.10	kJ/mol	Joback Method
hf	-2207.17	kJ/mol	Joback Method
hfus	179.30	kJ/mol	Joback Method
hvap	196.27	kJ/mol	Joback Method
log10ws	-25.40		Crippen Method
logp	23.450		Crippen Method
mcvol	1005.390	ml/mol	McGowan Method
pc	168.84	kPa	Joback Method
tb	2006.55	K	Joback Method
tf	1068.87	K	Joback Method
tt	340.60 ± 1.00	K	NIST Webbook
tt	346.50 ± 1.00	K	NIST Webbook
tt	355.50 ± 1.00	K	NIST Webbook
vc	3.966	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	4189.00	J/mol×K	2006.55	Joback Method
dvisc	0.0000002	Paxs	1068.87	Joback Method
dvisc	8.6508315e-08	Paxs	1225.15	Joback Method
dvisc	4.1518222e-08	Paxs	1381.43	Joback Method
dvisc	2.3132569e-08	Paxs	1537.71	Joback Method
dvisc	1.4357458e-08	Paxs	1693.99	Joback Method
dvisc	9.6588133e-09	Paxs	1850.27	Joback Method
dvisc	6.9117237e-09	Paxs	2006.55	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18641571&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18641571&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/17-718-6/Docosanoic-acid-1-2-3-propanetriyl-ester.pdf>

Generated by Cheméo on 2024-04-18 04:50:09.012347101 +0000 UTC m=+15705057.932924417.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.