

2,6-Difluoro-3-methylbenzoic acid, 4-benzyloxyphenyl ester

Inchi:	InChI=1S/C21H16F2O3/c1-14-7-12-18(22)19(20(14)23)21(24)26-17-10-8-16(9-11-17)25
InchiKey:	HWOHMPVKLMSDKK-UHFFFAOYSA-N
Formula:	C21H16F2O3
SMILES:	<chem>Cc1ccc(F)c(C(=O)Oc2ccc(OCc3ccccc3)cc2)c1F</chem>
Mol. weight [g/mol]:	354.35

Physical Properties

Property code	Value	Unit	Source
gf	-303.89	kJ/mol	Joback Method
hf	-582.30	kJ/mol	Joback Method
hfus	40.85	kJ/mol	Joback Method
hvap	81.75	kJ/mol	Joback Method
log10ws	-6.93		Crippen Method
logp	5.071		Crippen Method
mcvol	252.320	ml/mol	McGowan Method
pc	1792.42	kPa	Joback Method
rinpol	2810.00		NIST Webbook
tb	877.09	K	Joback Method
tc	1111.02	K	Joback Method
tf	551.34	K	Joback Method
vc	0.966	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	738.56	J/molxK	877.09	Joback Method
cpg	751.59	J/molxK	916.08	Joback Method
cpg	763.29	J/molxK	955.07	Joback Method
cpg	773.69	J/molxK	994.05	Joback Method
cpg	782.83	J/molxK	1033.04	Joback Method
cpg	790.75	J/molxK	1072.03	Joback Method
cpg	797.49	J/molxK	1111.02	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357688&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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