

# 2-(2-Ethoxyethoxy)ethyl nonanoate

<b>Inchi:</b>	InChI=1S/C15H30O4/c1-3-5-6-7-8-9-10-15(16)19-14-13-18-12-11-17-4-2/h3-14H2,1-2H3
<b>InchiKey:</b>	QNOCBHYYGWFCED-UHFFFAOYSA-N
<b>Formula:</b>	C15H30O4
<b>SMILES:</b>	CCCCCCCCC(=O)OCCOCCOCC
<b>Mol. weight [g/mol]:</b>	274.40

## Physical Properties

Property code	Value	Unit	Source
gf	-368.50	kJ/mol	Joback Method
hf	-862.17	kJ/mol	Joback Method
hfus	39.77	kJ/mol	Joback Method
hvap	62.96	kJ/mol	Joback Method
log10ws	-3.14		Crippen Method
logp	3.333		Crippen Method
mvol	241.390	ml/mol	McGowan Method
pc	1428.30	kPa	Joback Method
rinpol	1838.00		NIST Webbook
rinpol	1838.00		NIST Webbook
tb	663.73	K	Joback Method
tc	832.31	K	Joback Method
tf	375.43	K	Joback Method
vc	0.935	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	681.20	J/molxK	663.73	Joback Method
cpg	698.21	J/molxK	691.83	Joback Method
cpg	714.51	J/molxK	719.92	Joback Method
cpg	730.10	J/molxK	748.02	Joback Method
cpg	744.97	J/molxK	776.12	Joback Method
cpg	759.12	J/molxK	804.21	Joback Method
cpg	772.55	J/molxK	832.31	Joback Method
dvisc	0.0012296	Paxs	375.43	Joback Method

dvisc	0.0006071	Paxs	423.48	Joback Method
dvisc	0.0003461	Paxs	471.53	Joback Method
dvisc	0.0002189	Paxs	519.58	Joback Method
dvisc	0.0001496	Paxs	567.63	Joback Method
dvisc	0.0001085	Paxs	615.68	Joback Method
dvisc	0.0000825	Paxs	663.73	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378249&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378249&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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