

# 6-Phenylhexanoic acid

<b>Other names:</b>	Benzenehexanoic acid
<b>Inchi:</b>	InChI=1S/C12H16O2/c13-12(14)10-6-2-5-9-11-7-3-1-4-8-11/h1,3-4,7-8H,2,5-6,9-10H2,(H
<b>InchiKey:</b>	JTXZPQIXIXYMDY-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O2
<b>SMILES:</b>	O=C(O)CCCCC1ccccc1
<b>Mol. weight [g/mol]:</b>	192.25
<b>CAS:</b>	5581-75-9

## Physical Properties

Property code	Value	Unit	Source
gf	-103.17	kJ/mol	Joback Method
hf	-319.29	kJ/mol	Joback Method
hfus	26.56	kJ/mol	Joback Method
hvap	68.01	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	2.874		Crippen Method
mcvol	163.620	ml/mol	McGowan Method
pc	2600.00	kPa	Critical-Point Measurements for Phenylethanoic to 7-Phenylheptanoic Acids
tb	646.69	K	Joback Method
tc	841.89	K	Joback Method
tf	362.17	K	Joback Method
vc	0.625	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.96	J/molxK	646.69	Joback Method
cpg	437.59	J/molxK	679.22	Joback Method
cpg	449.48	J/molxK	711.76	Joback Method
cpg	460.66	J/molxK	744.29	Joback Method
cpg	471.16	J/molxK	776.82	Joback Method
cpg	481.02	J/molxK	809.36	Joback Method

cpg	490.27	J/molxK	841.89	Joback Method
dvisc	0.0043132	Paxs	362.17	Joback Method
dvisc	0.0014364	Paxs	409.59	Joback Method
dvisc	0.0006010	Paxs	457.01	Joback Method
dvisc	0.0002962	Paxs	504.43	Joback Method
dvisc	0.0001649	Paxs	551.85	Joback Method
dvisc	0.0001007	Paxs	599.27	Joback Method
dvisc	0.0000661	Paxs	646.69	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	474.70	K	3.20	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Critical-Point Measurements for Phenylethanoic to 7-Phenylheptanoic Acids</b>	<a href="https://www.doi.org/10.1021/je060078g">https://www.doi.org/10.1021/je060078g</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5581759&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5581759&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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