

# 4-Nitrobenzoic acid, oct-3-en-2-yl ester

<b>Inchi:</b>	InChI=1S/C15H19NO4/c1-3-4-5-6-7-12(2)20-15(17)13-8-10-14(11-9-13)16(18)19/h6-12H
<b>InchiKey:</b>	LXVKUUDHSMVOA-VOTSOKGWSA-N
<b>Formula:</b>	C15H19NO4
<b>SMILES:</b>	CCCCC=CC(C)OC(=O)c1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	277.32

## Physical Properties

Property code	Value	Unit	Source
gf	57.61	kJ/mol	Joback Method
hf	-271.49	kJ/mol	Joback Method
hfus	39.09	kJ/mol	Joback Method
hvap	77.24	kJ/mol	Joback Method
log10ws	-5.26		Crippen Method
logp	3.886		Crippen Method
mcvol	219.010	ml/mol	McGowan Method
pc	2021.76	kPa	Joback Method
rinsol	2023.00		NIST Webbook
tb	806.11	K	Joback Method
tc	1035.35	K	Joback Method
tf	493.44	K	Joback Method
vc	0.848	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	634.58	J/mol×K	806.11	Joback Method
cpg	648.57	J/mol×K	844.32	Joback Method
cpg	661.51	J/mol×K	882.52	Joback Method
cpg	673.46	J/mol×K	920.73	Joback Method
cpg	684.47	J/mol×K	958.94	Joback Method
cpg	694.61	J/mol×K	997.14	Joback Method
cpg	703.92	J/mol×K	1035.35	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299268&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299268&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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