

# D-(+)-Xylose, tetramethyl ether

<b>Other names:</b>	Methyl 2,3,4-tri-O-methyl-d-xylopyranoside
<b>Inchi:</b>	InChI=1S/C9H18O5/c1-10-6-5-14-9(13-4)8(12-3)7(6)11-2/h6-9H,5H2,1-4H3
<b>InchiKey:</b>	OQFUMNZJTHPYPI-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O5
<b>SMILES:</b>	COC1COC(OC)C(OC)C1OC
<b>Mol. weight [g/mol]:</b>	206.24

## Physical Properties

Property code	Value	Unit	Source
gf	-479.90	kJ/mol	Joback Method
hf	-896.67	kJ/mol	Joback Method
hfus	26.85	kJ/mol	Joback Method
hvap	49.28	kJ/mol	Joback Method
log10ws	0.14		Crippen Method
logp	0.034		Crippen Method
mcvol	156.160	ml/mol	McGowan Method
pc	2372.59	kPa	Joback Method
rinpol	1254.20		NIST Webbook
tb	527.49	K	Joback Method
tc	720.09	K	Joback Method
tf	301.34	K	Joback Method
vc	0.562	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	397.96	J/molxK	527.49	Joback Method
cpg	479.64	J/molxK	687.99	Joback Method
cpg	464.58	J/molxK	655.89	Joback Method
cpg	448.84	J/molxK	623.79	Joback Method
cpg	432.47	J/molxK	591.69	Joback Method
cpg	415.49	J/molxK	559.59	Joback Method
cpg	493.98	J/molxK	720.09	Joback Method
dvisc	0.0001730	Paxs	527.49	Joback Method

dvisc	0.0002078	Paxs	489.80	Joback Method
dvisc	0.0002572	Paxs	452.11	Joback Method
dvisc	0.0003310	Paxs	414.42	Joback Method
dvisc	0.0004481	Paxs	376.72	Joback Method
dvisc	0.0006489	Paxs	339.03	Joback Method
dvisc	0.0010307	Paxs	301.34	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U332893&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U332893&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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