

# Butanenitrile, 4,4-diethoxy-

<b>Other names:</b>	3-Cyanopropionaldehyde diethyl acetal Propionaldehyde, 3-cyano-, diethyl acetal 4,4-diethoxybutanenitrile
<b>Inchi:</b>	InChI=1S/C8H15NO2/c1-3-10-8(11-4-2)6-5-7-9/h8H,3-6H2,1-2H3
<b>InchiKey:</b>	DRZCPHGVEATLFR-UHFFFAOYSA-N
<b>Formula:</b>	C8H15NO2
<b>SMILES:</b>	CCOC(CCC#N)OCC
<b>Mol. weight [g/mol]:</b>	157.21
<b>CAS:</b>	18381-45-8

## Physical Properties

Property code	Value	Unit	Source
gf	-62.78	kJ/mol	Joback Method
hf	-313.29	kJ/mol	Joback Method
hfus	16.84	kJ/mol	Joback Method
hvap	48.31	kJ/mol	Joback Method
log10ws	-1.82		Crippen Method
logp	1.689		Crippen Method
mcvol	136.700	ml/mol	McGowan Method
pc	2410.00	kPa	Joback Method
tb	528.92	K	Joback Method
tc	715.81	K	Joback Method
tf	274.37	K	Joback Method
vc	0.539	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	317.72	J/molxK	528.92	Joback Method
cpg	329.35	J/molxK	560.07	Joback Method
cpg	340.56	J/molxK	591.22	Joback Method
cpg	351.34	J/molxK	622.37	Joback Method
cpg	361.70	J/molxK	653.51	Joback Method
cpg	371.61	J/molxK	684.66	Joback Method

cpg

381.09

J/mol×K

715.81

Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	378.20	K	1.30	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18381458&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18381458&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/17-828-4/Butanenitrile-4-4-diethoxy.pdf>

Generated by Cheméo on 2024-04-25 18:58:53.095914375 +0000 UTC m=+16360782.016491688.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.