

# cis-2-Methyl-2-butenedioic acid, dimethyl ester

<b>Other names:</b>	Citraconic acid, dimethyl ester Dimethyl citraconate Dimethyl methyl maleate Methylmaleic acid, dimethyl ester 2-Butenedioic acid, 2-methyl-, dimethyl ester (Z)- Maleic acid, methyl-, dimethyl ester 2-Methyl-but-2-enedioic acid dimethyl ester, Z
<b>Inchi:</b>	InChI=1S/C7H10O4/c1-5(7(9)11-3)4-6(8)10-2/h4H,1-3H3/b5-4-
<b>InchiKey:</b>	WQEXBUQDXKPVHR-PLNGDYQASA-N
<b>Formula:</b>	C7H10O4
<b>SMILES:</b>	COC(=O)C=C(C)C(=O)OC
<b>Mol. weight [g/mol]:</b>	158.15
<b>CAS:</b>	617-54-9

## Physical Properties

Property code	Value	Unit	Source
gf	-388.11	kJ/mol	Joback Method
hf	-569.98	kJ/mol	Joback Method
hfus	18.35	kJ/mol	Joback Method
hvap	49.53	kJ/mol	Joback Method
log10ws	-0.33		Crippen Method
logp	0.279		Crippen Method
mcvol	120.070	ml/mol	McGowan Method
pc	3299.15	kPa	Joback Method
rinpol	1058.00		NIST Webbook
rinpol	1055.00		NIST Webbook
rinpol	1059.00		NIST Webbook
rinpol	1045.00		NIST Webbook
rinpol	1058.00		NIST Webbook
rinpol	1056.00		NIST Webbook
tb	483.70	K	NIST Webbook
tc	713.77	K	Joback Method
tf	293.93	K	Joback Method
vc	0.457	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	305.95	J/mol×K	680.84	Joback Method
cpg	259.92	J/mol×K	516.18	Joback Method
cpg	269.97	J/mol×K	549.11	Joback Method
cpg	279.60	J/mol×K	582.04	Joback Method
cpg	288.81	J/mol×K	614.98	Joback Method
cpg	297.60	J/mol×K	647.91	Joback Method
cpg	313.89	J/mol×K	713.77	Joback Method
hvapt	55.80	kJ/mol	404.00	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	366.00	K	1.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C617549&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C617549&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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