

# N-Ethyl-p-toluidine

<b>Other names:</b>	Benzenamine, N-ethyl-4-methyl- N-Ethyl-4-methylaniline N-Ethyl-4-toluidine N-Ethyl-p-methylaniline NSC 8889 p-Methyl-N-ethylaniline p-Toluidine, N-ethyl-
<b>Inchi:</b>	InChI=1S/C9H13N/c1-3-10-9-6-4-8(2)5-7-9/h4-7,10H,3H2,1-2H3
<b>InchiKey:</b>	AASABFUMCBTXRL-UHFFFAOYSA-N
<b>Formula:</b>	C9H13N
<b>SMILES:</b>	CCNc1ccc(C)cc1
<b>Mol. weight [g/mol]:</b>	135.21
<b>CAS:</b>	622-57-1

## Physical Properties

Property code	Value	Unit	Source
gf	217.07	kJ/mol	Joback Method
hf	49.44	kJ/mol	Joback Method
hfus	17.82	kJ/mol	Joback Method
hvap	45.00	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	2.427		Crippen Method
mcvol	123.890	ml/mol	McGowan Method
pc	3254.14	kPa	Joback Method
rinpol	1276.40		NIST Webbook
rinpol	1277.30		NIST Webbook
ripol	1806.60		NIST Webbook
tb	490.20	K	NIST Webbook
tc	697.80	K	Joback Method
tf	282.79	K	Joback Method
vc	0.467	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	260.77	J/mol×K	487.15	Joback Method
cpg	274.66	J/mol×K	522.26	Joback Method
cpg	287.77	J/mol×K	557.37	Joback Method
cpg	300.15	J/mol×K	592.47	Joback Method
cpg	311.81	J/mol×K	627.58	Joback Method
cpg	322.79	J/mol×K	662.69	Joback Method
cpg	333.12	J/mol×K	697.80	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.48121e+01
Coeff. B	-4.20787e+03
Coeff. C	-7.74120e+01
Temperature range (K), min.	367.12
Temperature range (K), max.	520.32

## Sources

The Yaws Handbook of Vapor

Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C622571&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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