

3-Chloro-2-fluorobenzoic acid, 2,3,4,6-tetrachlorophenyl ester

Inchi: InChI=1S/C13H4Cl5FO2/c14-6-3-1-2-5(11(6)19)13(20)21-12-8(16)4-7(15)9(17)10(12)18
InchiKey: VWPZGMVLINDTQY-UHFFFAOYSA-N
Formula: C13H4Cl5FO2
SMILES: O=C(Oc1c(Cl)cc(Cl)c(Cl)c1Cl)c1cccc(Cl)c1F
Mol. weight [g/mol]: 388.43

Physical Properties

Property code	Value	Unit	Source
gf	-262.76	kJ/mol	Joback Method
hf	-427.02	kJ/mol	Joback Method
hfus	42.03	kJ/mol	Joback Method
hvap	83.32	kJ/mol	Joback Method
log10ws	-7.32		Crippen Method
logp	6.312		Crippen Method
mvol	216.920	ml/mol	McGowan Method
pc	2315.84	kPa	Joback Method
rinpol	2541.00		NIST Webbook
rinpol	2541.00		NIST Webbook
tb	842.79	K	Joback Method
tc	1094.87	K	Joback Method
tf	586.58	K	Joback Method
vc	0.835	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	467.45	J/molxK	842.79	Joback Method
cpg	474.84	J/molxK	884.80	Joback Method
cpg	481.35	J/molxK	926.82	Joback Method
cpg	487.01	J/molxK	968.83	Joback Method
cpg	491.82	J/molxK	1010.85	Joback Method
cpg	495.80	J/molxK	1052.86	Joback Method
cpg	498.96	J/molxK	1094.87	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U360589&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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