

# Trichloroacetic acid 2-propenyl ester

<b>Other names:</b>	Allyl trichloroacetate Acetic acid, trichloro-, allyl ester Acetic acid, trichloro-, 2-propenyl ester Trichloroacetic acid allyl ester
<b>Inchi:</b>	InChI=1S/C5H5Cl3O2/c1-2-3-10-4(9)5(6,7)8/h2H,1,3H2
<b>InchiKey:</b>	LJQCONXCOYBYIE-UHFFFAOYSA-N
<b>Formula:</b>	C5H5Cl3O2
<b>SMILES:</b>	C=CCOC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	203.45
<b>CAS:</b>	6304-34-3

## Physical Properties

Property code	Value	Unit	Source
chl	-2349.00	kJ/mol	NIST Webbook
chl	-2356.00 ± 4.00	kJ/mol	NIST Webbook
gf	-187.81	kJ/mol	Joback Method
hf	-343.00 ± 9.60	kJ/mol	NIST Webbook
hfl	-395.00 ± 8.40	kJ/mol	NIST Webbook
hfus	15.39	kJ/mol	Joback Method
hvap	52.30 ± 4.20	kJ/mol	NIST Webbook
log10ws	-2.19		Crippen Method
logp	2.086		Crippen Method
mcvol	121.170	ml/mol	McGowan Method
pc	3452.08	kPa	Joback Method
rinpol	1044.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	1027.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	1027.00		NIST Webbook
ripol	1467.00		NIST Webbook
ripol	1467.00		NIST Webbook
tb	495.83	K	Joback Method
tc	711.57	K	Joback Method
tf	308.69	K	Joback Method
vc	0.457	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	217.41	J/molxK	495.83	Joback Method
cpg	249.72	J/molxK	675.61	Joback Method
cpg	244.25	J/molxK	639.66	Joback Method
cpg	238.31	J/molxK	603.70	Joback Method
cpg	231.88	J/molxK	567.74	Joback Method
cpg	224.92	J/molxK	531.79	Joback Method
cpg	254.75	J/molxK	711.57	Joback Method
dvisc	0.0003259	Paxs	495.83	Joback Method
dvisc	0.0004180	Paxs	464.64	Joback Method
dvisc	0.0005558	Paxs	433.45	Joback Method
dvisc	0.0007722	Paxs	402.26	Joback Method
dvisc	0.0011340	Paxs	371.07	Joback Method
dvisc	0.0017869	Paxs	339.88	Joback Method
dvisc	0.0030869	Paxs	308.69	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6304343&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6304343&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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