

# 1,3,5-Triphenyl-1,5-pentadione

**Inchi:** InChI=1S/C23H20O2/c24-22(19-12-6-2-7-13-19)16-21(18-10-4-1-5-11-18)17-23(25)20-1

**InchiKey:** SQFIJFCIFAVGEG-UHFFFAOYSA-N

**Formula:** C23H20O2

**SMILES:** O=C(CC(CC(=O)c1ccccc1)c1ccccc1)c1ccccc1

**Mol. weight [g/mol]:** 328.41

## Physical Properties

| Property code | Value   | Unit    | Source             |
|---------------|---------|---------|--------------------|
| af            | 0.6858  |         | Cheméo Relay (1.0) |
| gf            | 219.73  | kJ/mol  | Joback Method      |
| hf            | -24.16  | kJ/mol  | Cheméo Relay (1.0) |
| hfus          | 37.12   | kJ/mol  | Joback Method      |
| hvap          | 123.18  | kJ/mol  | Cheméo Relay (1.0) |
| ie            | 8.69    | eV      | Cheméo Relay (1.0) |
| log10ws       | -5.40   |         | Cheméo Relay (1.0) |
| logp          | 5.316   |         | Crippen Method     |
| mcvol         | 266.790 | ml/mol  | McGowan Method     |
| pc            | 1892.00 | kPa     | Joback Method      |
| tb            | 672.76  | K       | Cheméo Relay (1.0) |
| tc            | 1004.60 | K       | Cheméo Relay (1.0) |
| tf            | 393.55  | K       | Cheméo Relay (1.0) |
| vc            | 0.944   | m3/kmol | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 801.04    | J/mol×K | 912.98          | Joback Method |
| cpg           | 815.32    | J/mol×K | 955.38          | Joback Method |
| cpg           | 828.28    | J/mol×K | 997.79          | Joback Method |
| cpg           | 840.06    | J/mol×K | 1040.19         | Joback Method |
| cpg           | 850.81    | J/mol×K | 1082.59         | Joback Method |
| cpg           | 860.67    | J/mol×K | 1125.00         | Joback Method |
| cpg           | 869.79    | J/mol×K | 1167.40         | Joback Method |
| dvisc         | 0.0008604 | Pa×s    | 513.09          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0004290 | Pa×s | 579.74 | Joback Method |
| dvisc | 0.0002469 | Pa×s | 646.39 | Joback Method |
| dvisc | 0.0001575 | Pa×s | 713.03 | Joback Method |
| dvisc | 0.0001085 | Pa×s | 779.68 | Joback Method |
| dvisc | 0.0000793 | Pa×s | 846.33 | Joback Method |
| dvisc | 0.0000607 | Pa×s | 912.98 | Joback Method |

## Sources

|                                  |                                                                                                                                             |
|----------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b>           | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| <b>Crippen Method:</b>           | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |
| <b>Cheméo Relay (1.0):</b>       |                                                                                                                                             |
| <b>Cheméo Relay (1.0, beta):</b> |                                                                                                                                             |
| <b>NIST Webbook:</b>             | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6263849&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6263849&amp;Units=SI</a> |
| <b>Joback Method:</b>            | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |
| <b>McGowan Method:</b>           | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>af:</b>      | Acentric Factor                                 |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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