

# Acetic acid, [3-hydroxy-4,4'-carbonylbis(phenyleneoxy)di-], didodecyl ester

InChI: InChI=1S/C41H62O8/c1-3-5-7-9-11-13-15-17-19-21-29-46-39(43)32-48-35-25-23-34(24-25)  
InChIKey: JIVKFMXHZGNYIX-UHFFFAOYSA-N

Formula: C41H62O8

SMILES: CCCCCCCCCCOC(=O)COc1ccc(C(=O)c2ccc(OCC(=O)OCCCCCCCCCCCC)cc2O)O

Mol. weight [g/mol]: 682.93

CAS: 116436-40-9

## Physical Properties

Property code	Value	Unit	Source
gf	-461.48	kJ/mol	Joback Method
hf	-1483.38	kJ/mol	Joback Method
hfus	104.58	kJ/mol	Joback Method
hvap	155.63	kJ/mol	Joback Method
log10ws	-11.83		Crippen Method
logp	10.309		Crippen Method
mcvol	575.090	ml/mol	McGowan Method
pc	553.11	kPa	Joback Method
tb	1532.71	K	Joback Method
tc	2172.05	K	Joback Method
tf	980.14	K	Joback Method
vc	2.171	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2184.16	J/molxK	1532.71	Joback Method
cpg	2204.84	J/molxK	1639.27	Joback Method
cpg	2223.29	J/molxK	1745.82	Joback Method
cpg	2242.17	J/molxK	1852.38	Joback Method
cpg	2264.12	J/molxK	1958.94	Joback Method
cpg	2291.80	J/molxK	2065.50	Joback Method
cpg	2327.84	J/molxK	2172.05	Joback Method
dvisc	0.0000003	Paxs	980.14	Joback Method
dvisc	0.0000001	Paxs	1072.24	Joback Method

dvisc	7.6343551e-08	Paxs	1164.33	Joback Method
dvisc	4.6501234e-08	Paxs	1256.42	Joback Method
dvisc	3.0308525e-08	Paxs	1348.52	Joback Method
dvisc	2.0865719e-08	Paxs	1440.61	Joback Method
dvisc	1.5024004e-08	Paxs	1532.71	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116436409&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116436409&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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