

Methyl 4-chloro-2,2-dimethyl-4-pentenoate

Inchi:	InChI=1S/C8H13ClO2/c1-6(9)5-8(2,3)7(10)11-4/h1,5H2,2-4H3
InchiKey:	NMIYZQVNXZOOPE-UHFFFAOYSA-N
Formula:	C8H13ClO2
SMILES:	C=C(Cl)CC(C)(C)C(=O)OC
Mol. weight [g/mol]:	176.64
CAS:	86799-85-1

Physical Properties

Property code	Value	Unit	Source
gf	-147.24	kJ/mol	Joback Method
hf	-362.10	kJ/mol	Joback Method
hfus	13.46	kJ/mol	Joback Method
hvap	45.06	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	2.328		Crippen Method
mcvol	138.960	ml/mol	McGowan Method
pc	2738.28	kPa	Joback Method
tb	489.49	K	Joback Method
tc	689.10	K	Joback Method
tf	268.70	K	Joback Method
vc	0.527	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	293.06	J/mol×K	489.49	Joback Method
cpg	305.45	J/mol×K	522.76	Joback Method
cpg	317.15	J/mol×K	556.03	Joback Method
cpg	328.19	J/mol×K	589.29	Joback Method
cpg	338.60	J/mol×K	622.56	Joback Method
cpg	348.40	J/mol×K	655.83	Joback Method
cpg	357.61	J/mol×K	689.10	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C86799851&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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