

# 3-((1,1-Dimethylethyl)-thio)-1-propene

<b>Other names:</b>	Allyl tert-butyl sulphide Allyl sec-butyl sulfide 5,5-Dimethyl-4-thia-1-hexene
<b>Inchi:</b>	InChI=1S/C7H14S/c1-5-6-8-7(2,3)4/h5H,1,6H2,2-4H3
<b>InchiKey:</b>	XIBMCVKDUXKQRS-UHFFFAOYSA-N
<b>Formula:</b>	C7H14S
<b>SMILES:</b>	C=CCSC(C)(C)C
<b>Mol. weight [g/mol]:</b>	130.25
<b>CAS:</b>	37850-75-2

## Physical Properties

Property code	Value	Unit	Source
chl	-5266.60 ± 2.50	kJ/mol	NIST Webbook
gf	131.86	kJ/mol	Joback Method
hf	-46.00 ± 2.00	kJ/mol	NIST Webbook
hfl	-90.70 ± 1.60	kJ/mol	NIST Webbook
hfus	9.32	kJ/mol	Joback Method
hvap	44.70	kJ/mol	NIST Webbook
hvap	44.80	kJ/mol	NIST Webbook
hvap	44.00 ± 1.00	kJ/mol	NIST Webbook
log10ws	-2.60		Crippen Method
logp	2.704		Crippen Method
mcvol	121.540	ml/mol	McGowan Method
pc	3045.68	kPa	Joback Method
rinpol	855.70		NIST Webbook
rinpol	877.00		NIST Webbook
rinpol	877.00		NIST Webbook
rinpol	877.00		NIST Webbook
tb	421.79	K	Joback Method
tc	626.22	K	Joback Method
tf	203.71	K	Joback Method
vc	0.452	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.80	J/mol×K	558.08	Joback Method
cpg	292.48	J/mol×K	592.15	Joback Method
cpg	232.08	J/mol×K	421.79	Joback Method
cpg	245.62	J/mol×K	455.86	Joback Method
cpg	258.40	J/mol×K	489.93	Joback Method
cpg	270.45	J/mol×K	524.01	Joback Method
cpg	302.53	J/mol×K	626.22	Joback Method
hvapt	41.90	kJ/mol	329.00	NIST Webbook
hvapt	43.10	kJ/mol	329.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37850752&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37850752&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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