

# Chloroxylonol

<b>Other names:</b>	2-Chloro-5-hydroxy-1,3-dimethylbenzene 2-Chloro-5-hydroxy-m-xylene 2-Chloro-m-xylonol 3,5-Dimethyl-4-chlorophenol 3,5-Xylonol, 4-chloro- 4-Chloro-1-hydroxy-3,5-dimethylbenzene 4-Chloro-3,5-dimethylphenol 4-Chloro-3,5-xylonol 4-Chloro-m-xylonol Ayrtol Benzytol Camel Desson Dettol Dettol, liquid antiseptic Espadol Husept extra NSC 4971 Nipacide PX Nipacide mx Ottasept Ottasept extra PCMX Para-chloro-meta-xylonol Parametaxylonol Phenol, 4-chloro-3,5-dimethyl- RBA 777 Septiderm-hydrochloride Willenol V m-Xylonol, 4-chloro- p-Chloro-3,5-dimethylphenol p-Chloro-3,5-xylonol p-Chloro-m-xylonol
<b>Inchi:</b>	InChI=1S/C8H9ClO/c1-5-3-7(10)4-6(2)8(5)9/h3-4,10H,1-2H3
<b>InchiKey:</b>	OSDLLIBGSJNGJE-UHFFFAOYSA-N
<b>Formula:</b>	C8H9ClO
<b>SMILES:</b>	<chem>Cc1cc(O)cc(C)c1Cl</chem>
<b>Mol. weight [g/mol]:</b>	156.61
<b>CAS:</b>	88-04-0

# Physical Properties

Property code	Value	Unit	Source
gf	-56.92	kJ/mol	Joback Method
hf	-187.91	kJ/mol	Joback Method
h <sub>fus</sub>	19.72	kJ/mol	Joback Method
h <sub>vap</sub>	54.40	kJ/mol	Joback Method
log <sub>10</sub> ws	-2.80		Aqueous Solubility Prediction Method
logp	2.662		Crippen Method
m <sub>cvol</sub>	117.930	ml/mol	McGowan Method
pc	4041.50	kPa	Joback Method
r <sub>inpol</sub>	1384.00		NIST Webbook
r <sub>inpol</sub>	1384.00		NIST Webbook
r <sub>ipol</sub>	2561.00		NIST Webbook
tb	537.13	K	Joback Method
tc	770.68	K	Joback Method
tf	387.90	K	Aqueous Solubility Prediction Method
vc	0.391	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	247.37	J/mol×K	537.13	Joback Method
c <sub>pg</sub>	257.62	J/mol×K	576.06	Joback Method
c <sub>pg</sub>	267.16	J/mol×K	614.98	Joback Method
c <sub>pg</sub>	276.06	J/mol×K	653.91	Joback Method
c <sub>pg</sub>	284.37	J/mol×K	692.83	Joback Method
c <sub>pg</sub>	292.19	J/mol×K	731.76	Joback Method
c <sub>pg</sub>	299.57	J/mol×K	770.68	Joback Method
d <sub>visc</sub>	0.0013492	Paxs	373.02	Joback Method
d <sub>visc</sub>	0.0006878	Paxs	400.37	Joback Method
d <sub>visc</sub>	0.0003822	Paxs	427.72	Joback Method
d <sub>visc</sub>	0.0002279	Paxs	455.07	Joback Method
d <sub>visc</sub>	0.0001441	Paxs	482.43	Joback Method
d <sub>visc</sub>	0.0000957	Paxs	509.78	Joback Method
d <sub>visc</sub>	0.0000663	Paxs	537.13	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C88040&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C88040&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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