

# 1,4-bromoethoxybutane

<b>Other names:</b>	Butane, 1-bromo-4-ethoxy
<b>Inchi:</b>	InChI=1S/C6H13BrO/c1-2-8-6-4-3-5-7/h2-6H2,1H3
<b>InchiKey:</b>	AABRBNAXKRILNN-UHFFFAOYSA-N
<b>Formula:</b>	C6H13BrO
<b>SMILES:</b>	CCOCCCCBr
<b>Mol. weight [g/mol]:</b>	181.07

## Physical Properties

Property code	Value	Unit	Source
gf	-91.04	kJ/mol	Joback Method
hf	-273.06	kJ/mol	Joback Method
hfus	17.77	kJ/mol	Joback Method
hvap	37.80	kJ/mol	Joback Method
log10ws	-1.85		Crippen Method
logp	2.198		Crippen Method
mcpol	118.770	ml/mol	McGowan Method
pc	3295.37	kPa	Joback Method
rinpol	994.30		NIST Webbook
rinpol	1001.30		NIST Webbook
rinpol	994.30		NIST Webbook
tb	425.26	K	Joback Method
tc	608.72	K	Joback Method
tf	239.41	K	Joback Method
vc	0.452	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.99	J/molxK	425.26	Joback Method
cpg	230.39	J/molxK	455.84	Joback Method
cpg	240.40	J/molxK	486.41	Joback Method
cpg	250.02	J/molxK	516.99	Joback Method
cpg	259.27	J/molxK	547.57	Joback Method
cpg	268.15	J/molxK	578.15	Joback Method

cpg	276.66	J/mol×K	608.72	Joback Method
dvisc	0.0030729	Paxs	239.41	Joback Method
dvisc	0.0016637	Paxs	270.38	Joback Method
dvisc	0.0010218	Paxs	301.36	Joback Method
dvisc	0.0006873	Paxs	332.33	Joback Method
dvisc	0.0004946	Paxs	363.31	Joback Method
dvisc	0.0003749	Paxs	394.28	Joback Method
dvisc	0.0002958	Paxs	425.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R135805&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R135805&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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