

# Acetic acid, [2-[(2-propenylamino)carbonyl]phenoxy]-

Other names:

Acetic acid, (o-(allylcarbamoyl)phenoxy)-

Acetic acid, (o-(N-allylcarbamoyl)phenoxy)-

Benzamide, N-allyl-, o-(carboxymethoxy)-

[2-[(allylamino)carbonyl]phenoxy]acetic acid

Inchi:

InChI=1S/C12H13NO4/c1-2-7-13-12(16)9-5-3-4-6-10(9)17-8-11(14)15/h2-6H,1,7-8H2,(H

InchiKey:

IHBWXJXQXPFAE-UHFFFAOYSA-N

Formula:

C12H13NO4

SMILES:

C=CCNC(=O)c1cccc1OCC(=O)O

Mol. weight [g/mol]:

235.24

CAS:

119-45-9

## Physical Properties

Property code	Value	Unit	Source
gf	-169.49	kJ/mol	Joback Method
hf	-396.66	kJ/mol	Joback Method
hfus	32.78	kJ/mol	Joback Method
hvap	83.59	kJ/mol	Joback Method
log10ws	-2.14		Crippen Method
logp	1.066		Crippen Method
mcvol	176.740	ml/mol	McGowan Method
pc	3093.29	kPa	Joback Method
tb	774.81	K	Joback Method
tc	980.36	K	Joback Method
tf	497.75	K	Joback Method
vc	0.664	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	487.16	J/molxK	774.81	Joback Method
cpg	497.20	J/molxK	809.07	Joback Method
cpg	506.51	J/molxK	843.33	Joback Method
cpg	515.12	J/molxK	877.58	Joback Method
cpg	523.04	J/molxK	911.84	Joback Method

cpg	530.30	J/mol×K	946.10	Joback Method
cpg	536.92	J/mol×K	980.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119459&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119459&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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