

# Pentacosanoic acid, methyl ester

<b>Other names:</b>	methyl pentacosanoate
<b>Inchi:</b>	InChI=1S/C26H52O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23-24-25
<b>InchiKey:</b>	WOPKHAQDUMDJY-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>26</sub> H <sub>52</sub> O <sub>2</sub>
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCCCCCCCC(=O)OC
<b>Mol. weight [g/mol]:</b>	396.69
<b>CAS:</b>	55373-89-2

## Physical Properties

Property code	Value	Unit	Source
gf	-65.88	kJ/mol	Joback Method
hf	-824.77	kJ/mol	Joback Method
hfus	65.88	kJ/mol	Joback Method
hvap	142.00 ± 4.50	kJ/mol	NIST Webbook
log10ws	-9.57		Crippen Method
logp	9.152		Crippen Method
mcvol	384.640	ml/mol	McGowan Method
pc	743.26	kPa	Joback Method
rinpol	2813.00		NIST Webbook
rinpol	464.30		NIST Webbook
rinpol	2822.90		NIST Webbook
rinpol	2813.71		NIST Webbook
rinpol	2817.00		NIST Webbook
rinpol	464.30		NIST Webbook
tb	870.57	K	Joback Method
tc	1067.44	K	Joback Method
tf	454.94	K	Joback Method
vc	1.516	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1403.78	J/mol×K	1067.44	Joback Method
cpg	1285.37	J/mol×K	870.57	Joback Method

cpg	1308.31	J/molxK	903.38	Joback Method
cpg	1329.90	J/molxK	936.19	Joback Method
cpg	1350.19	J/molxK	969.01	Joback Method
cpg	1369.24	J/molxK	1001.82	Joback Method
cpg	1387.08	J/molxK	1034.63	Joback Method
dvisc	0.0000311	Paxs	870.57	Joback Method
dvisc	0.0008279	Paxs	454.94	Joback Method
dvisc	0.0003338	Paxs	524.21	Joback Method
dvisc	0.0001664	Paxs	593.48	Joback Method
dvisc	0.0000959	Paxs	662.75	Joback Method
dvisc	0.0000614	Paxs	732.03	Joback Method
dvisc	0.0000424	Paxs	801.30	Joback Method
hfust	92.00	kJ/mol	332.20	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55373892&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55373892&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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