

Ethyl iodoacetate

Other names:	Acetic acid, iodo-, ethyl ester Ethyl monoiodoacetate Iodoacetic acid, ethyl ester S 9
Inchi:	InChI=1S/C4H7IO2/c1-2-7-4(6)3-5/h2-3H2,1H3
InchiKey:	MFFXVVHUKRKXCI-UHFFFAOYSA-N
Formula:	C4H7IO2
SMILES:	CCOC(=O)CI
Mol. weight [g/mol]:	214.00
CAS:	623-48-3

Physical Properties

Property code	Value	Unit	Source
gf	-193.00	kJ/mol	Joback Method
hf	-293.82	kJ/mol	Joback Method
hfus	13.31	kJ/mol	Joback Method
hvap	43.03	kJ/mol	Joback Method
log10ws	-1.31		Crippen Method
logp	0.984		Crippen Method
mcvol	100.480	ml/mol	McGowan Method
pc	4015.93	kPa	Joback Method
rinpola	960.50		NIST Webbook
tb	452.20	K	NIST Webbook
tc	676.34	K	Joback Method
tf	265.06	K	Joback Method
vc	0.371	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	205.39	J/mol×K	676.34	Joback Method
cpg	199.76	J/mol×K	640.34	Joback Method
cpg	193.82	J/mol×K	604.35	Joback Method
cpg	187.57	J/mol×K	568.35	Joback Method

cpg	180.99	J/molxK	532.35	Joback Method
cpg	174.08	J/molxK	496.35	Joback Method
cpg	166.84	J/molxK	460.35	Joback Method
dvisc	0.0037331	Paxs	265.06	Joback Method
dvisc	0.0003904	Paxs	460.35	Joback Method
dvisc	0.0004929	Paxs	427.80	Joback Method
dvisc	0.0006466	Paxs	395.25	Joback Method
dvisc	0.0008908	Paxs	362.71	Joback Method
dvisc	0.0013071	Paxs	330.16	Joback Method
dvisc	0.0020858	Paxs	297.61	Joback Method
hvapt	52.10	kJ/mol	331.50	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	346.20	K	2.10	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.84853e+01
Coeff. B	-6.26994e+03
Temperature range (K), min.	344.55
Temperature range (K), max.	475.94

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C623483&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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