

# 1,3-Dithiolane

<b>Other names:</b>	1,3-Dithiacyclopentane
<b>Inchi:</b>	InChI=1S/C3H6S2/c1-2-5-3-4-1/h1-3H2
<b>InchiKey:</b>	IMLSAISZLJGWPP-UHFFFAOYSA-N
<b>Formula:</b>	C3H6S2
<b>SMILES:</b>	C1CSCS1
<b>Mol. weight [g/mol]:</b>	106.21
<b>CAS:</b>	4829-04-3

## Physical Properties

Property code	Value	Unit	Source
gf	98.36	kJ/mol	Joback Method
hf	66.09	kJ/mol	Joback Method
hfus	3.70	kJ/mol	Joback Method
hvap	34.46	kJ/mol	Joback Method
ie	8.75	eV	NIST Webbook
ie	8.77	eV	NIST Webbook
ie	9.00 ± 0.05	eV	NIST Webbook
log10ws	-1.24		Crippen Method
logp	1.424		Crippen Method
mcvol	74.970	ml/mol	McGowan Method
pc	5871.90	kPa	Joback Method
rinpol	1036.00		NIST Webbook
rinpol	1018.00		NIST Webbook
rinpol	1013.00		NIST Webbook
rinpol	1006.00		NIST Webbook
rinpol	1014.00		NIST Webbook
rinpol	1014.00		NIST Webbook
rinpol	1006.00		NIST Webbook
rinpol	994.00		NIST Webbook
rinpol	978.00		NIST Webbook
tb	448.20	K	NIST Webbook
tc	623.34	K	Joback Method
tf	305.61	K	Joback Method
vc	0.237	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	113.20	J/mol×K	383.65	Joback Method
cpg	122.76	J/mol×K	423.60	Joback Method
cpg	131.61	J/mol×K	463.55	Joback Method
cpg	139.81	J/mol×K	503.50	Joback Method
cpg	147.38	J/mol×K	543.44	Joback Method
cpg	154.38	J/mol×K	583.39	Joback Method
cpg	160.85	J/mol×K	623.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4829043&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4829043&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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