

# Benzeneacetonitrile, «alpha»-methyl-

<b>Other names:</b>	«alpha»-Cyanoethylbenzene Hydratroponitrile «alpha»-Methylbenzeneacetonitrile «alpha»-Methylbenzyl cyanide «alpha»-Phenylethyl cyanide 2-Phenylpropanenitrile «alpha»-Phenylpropionitrile 2-Phenylpropionitrile «alpha»-Methylphenylacetonitrile 2-Phenylpropyl nitrile 2-phenylpropionitrile
<b>Inchi:</b>	InChI=1S/C9H9N/c1-8(7-10)9-5-3-2-4-6-9/h2-6,8H,1H3
<b>InchiKey:</b>	NVAOLENBKNECGF-UHFFFAOYSA-N
<b>Formula:</b>	C9H9N
<b>SMILES:</b>	CC(C#N)c1ccccc1
<b>Mol. weight [g/mol]:</b>	131.17
<b>CAS:</b>	1823-91-2

## Physical Properties

Property code	Value	Unit	Source
gf	268.05	kJ/mol	Joback Method
hf	167.04	kJ/mol	Joback Method
hfus	11.09	kJ/mol	Joback Method
hvap	60.90 ± 0.70	kJ/mol	NIST Webbook
log10ws	-2.52		Crippen Method
logp	2.314		Crippen Method
mcvol	115.290	ml/mol	McGowan Method
pc	3199.16	kPa	Joback Method
rinpol	1227.00		NIST Webbook
ripol	2038.00		NIST Webbook
tb	533.64	K	Joback Method
tc	766.36	K	Joback Method
tf	267.60	K	Joback Method
vc	0.452	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	296.19	J/mol×K	727.58	Joback Method
cpg	244.47	J/mol×K	533.64	Joback Method
cpg	256.37	J/mol×K	572.43	Joback Method
cpg	267.46	J/mol×K	611.21	Joback Method
cpg	277.76	J/mol×K	650.00	Joback Method
cpg	287.32	J/mol×K	688.79	Joback Method
cpg	304.39	J/mol×K	766.36	Joback Method
hvapt	60.80 ± 0.70	kJ/mol	301.00	NIST Webbook

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1823912&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1823912&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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