

# Benzoic acid, 2-(isopropyl)oxy-, methyl ester

Inchi:	InChI=1S/C11H14O3/c1-8(2)14-10-7-5-4-6-9(10)11(12)13-3/h4-8H,1-3H3
InchiKey:	RDAALIJZJHPRRP-UHFFFAOYSA-N
Formula:	C11H14O3
SMILES:	COC(=O)c1ccccc1OC(C)C
Mol. weight [g/mol]:	194.23

## Physical Properties

Property code	Value	Unit	Source
gf	-196.84	kJ/mol	Joback Method
hf	-427.61	kJ/mol	Joback Method
hfus	18.35	kJ/mol	Joback Method
hvap	54.20	kJ/mol	Joback Method
log10ws	-2.78		Crippen Method
logp	2.260		Crippen Method
mcvol	155.400	ml/mol	McGowan Method
pc	2709.85	kPa	Joback Method
rinpol	1409.00		NIST Webbook
tb	581.01	K	Joback Method
tc	793.47	K	Joback Method
tf	332.06	K	Joback Method
vc	0.580	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.20	J/molxK	581.01	Joback Method
cpg	385.45	J/molxK	616.42	Joback Method
cpg	398.95	J/molxK	651.83	Joback Method
cpg	411.70	J/molxK	687.24	Joback Method
cpg	423.70	J/molxK	722.65	Joback Method
cpg	434.96	J/molxK	758.06	Joback Method
cpg	445.46	J/molxK	793.47	Joback Method
dvisc	0.0016320	Paxs	332.06	Joback Method
dvisc	0.0008770	Paxs	373.55	Joback Method

dvisc	0.0005336	Paxs	415.04	Joback Method
dvisc	0.0003554	Paxs	456.53	Joback Method
dvisc	0.0002532	Paxs	498.03	Joback Method
dvisc	0.0001901	Paxs	539.52	Joback Method
dvisc	0.0001487	Paxs	581.01	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375335&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375335&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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