

# 1-Benzylpyrene

<b>Inchi:</b>	InChI=1S/C23H16/c1-2-5-16(6-3-1)15-20-12-11-19-10-9-17-7-4-8-18-13-14-21(20)23(19)
<b>InchiKey:</b>	KBUVRXNOTZNAPI-UHFFFAOYSA-N
<b>Formula:</b>	C23H16
<b>SMILES:</b>	c1ccc(Cc2ccc3ccc4cccc5ccc2c3c45)cc1
<b>Mol. weight [g/mol]:</b>	292.37
<b>CAS:</b>	42211-34-7

## Physical Properties

Property code	Value	Unit	Source
gf	652.90	kJ/mol	Joback Method
hf	448.35	kJ/mol	Joback Method
hfus	36.28	kJ/mol	Joback Method
hvap	77.61	kJ/mol	Joback Method
log10ws	-8.40		Crippen Method
logp	6.175		Crippen Method
mcvol	233.330	ml/mol	McGowan Method
pc	2143.35	kPa	Joback Method
tb	843.18	K	Joback Method
tc	1106.73	K	Joback Method
tf	362.15 ± 2.00	K	NIST Webbook
vc	0.903	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	662.95	J/mol×K	843.18	Joback Method
cpg	678.24	J/mol×K	887.11	Joback Method
cpg	692.87	J/mol×K	931.03	Joback Method
cpg	707.10	J/mol×K	974.96	Joback Method
cpg	721.21	J/mol×K	1018.88	Joback Method
cpg	735.47	J/mol×K	1062.81	Joback Method
cpg	750.15	J/mol×K	1106.73	Joback Method
dvisc	0.0022173	Paxs	543.75	Joback Method
dvisc	0.0018517	Paxs	593.65	Joback Method

dvisc	0.0015902	Paxs	643.56	Joback Method
dvisc	0.0013959	Paxs	693.47	Joback Method
dvisc	0.0012469	Paxs	743.37	Joback Method
dvisc	0.0011298	Paxs	793.28	Joback Method
dvisc	0.0010357	Paxs	843.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42211347&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42211347&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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