

# Dimethylmalonic acid, 2,5-dichlorophenyl ethyl ester

<b>Inchi:</b>	InChI=1S/C13H14Cl2O4/c1-4-18-11(16)13(2,3)12(17)19-10-7-8(14)5-6-9(10)15/h5-7H,4
<b>InchiKey:</b>	PCRJTCKWWJXKBD-UHFFFAOYSA-N
<b>Formula:</b>	C13H14Cl2O4
<b>SMILES:</b>	CCOC(=O)C(C)(C)C(=O)Oc1cc(Cl)ccc1Cl
<b>Mol. weight [g/mol]:</b>	305.15

## Physical Properties

Property code	Value	Unit	Source
gf	-337.13	kJ/mol	Joback Method
hf	-627.89	kJ/mol	Joback Method
hfus	29.24	kJ/mol	Joback Method
hvap	73.92	kJ/mol	Joback Method
log10ws	-3.87		Crippen Method
logp	3.488		Crippen Method
mvol	209.630	ml/mol	McGowan Method
pc	2208.29	kPa	Joback Method
rinpol	1863.00		NIST Webbook
rinpol	1863.00		NIST Webbook
tb	757.69	K	Joback Method
tc	984.10	K	Joback Method
tf	494.31	K	Joback Method
vc	0.790	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	537.95	J/molxK	757.69	Joback Method
cpg	549.88	J/molxK	795.42	Joback Method
cpg	560.86	J/molxK	833.16	Joback Method
cpg	570.89	J/molxK	870.89	Joback Method
cpg	580.00	J/molxK	908.63	Joback Method
cpg	588.23	J/molxK	946.36	Joback Method
cpg	595.60	J/molxK	984.10	Joback Method
dvisc	0.0006326	Paxs	494.31	Joback Method

dvisc	0.0003995	Paxs	538.21	Joback Method
dvisc	0.0002704	Paxs	582.10	Joback Method
dvisc	0.0001933	Paxs	626.00	Joback Method
dvisc	0.0001444	Paxs	669.90	Joback Method
dvisc	0.0001119	Paxs	713.79	Joback Method
dvisc	0.0000892	Paxs	757.69	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363677&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363677&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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